SEARCH REQUEST FORM

Scientific and Technical Information Center

Gelekti			
Paguester's Full Name: 10 100 Ch	Exa	miner #: Date: 3/0/04 Serial Number: 10/00/756 Demait Preferred (circles: PAPER DISK E-MAIL	
Art Unit: 17() Phone Numb	er-30 <u>(7/. / 57 9</u>	Serial Number: 10/00/05/Cormat Preferred (circle): PAPER DISK E-MAIL	
If more than one search is submitted	, please prioritize se *******	arches in order of need. ***********************************	
Please provide a detailed statement of the search	topic, and describe as spe	and expired numbers, and combine with the concept or	
utility of the invention. Define any terms that n known. Please attach a copy of the cover sheet,	lay have a special mount	•	
Title of Invention: \(\text{hotopotler} \)	nable MILLEC	War Circuitry	
Inventors (please provide full names): Var	rica A So	ck Xao-An Thong,	
Zhang - Lin Zhou Earliest Priority Filing Date:			
Earliest Priority Filing Date:	-	alone with the	
For Sequence Searches Only Please include all appropriate serial number.	pertinent information (parer	ıt, child, divisional, or issued patent numbers) along with the	
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nothing foundher search	the tombor	ands in claims 13,14 and	
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STAFF USE ONLY	Type of Search	Vendors and cost where applicable	
Searcher:	NA Sequence (#)	STN	· newspaper of the Solding
Searcher Phone #:	AA Sequence (#)	DialogQuestel/Orbit	
Searcher Location:	Structure (#)	Questel/Orbit Dr.Link	
Date Searcher Picked Up:	Bibliographic	Lexis/Nexis	
Date Completed:	Litigation	Sequence Systems	
Searcher Prep & Review Time:	Fulltext	WWW/Internet	
Clerical Prep Time:	Patent Family	Other (specify)	?
Online Time:		** %	
PTO-1590 (1-2000)	プイト アイルニー		

ANSWER 1 OF 1 HCAPLUS COPYRIGHT 2004 ACS on STN L12003:818009 HCAPLUS ΝA 139:314469 DN

Entered STN: 17 Oct 2003 ED

Photopatternable molecular circuitry TI

Beck, Patricia A.; Zhang, Xiao-An; Zhou, Zhang-Lin IN

PA

U.S. Pat. Appl. Publ., 39 pp. CODEN: USXXCO

Patent DT

English LA

ICM G03F007-004 IC ICS G03C001-52

430146000; 430171000; 430270100; 430311000 NCL

applicant

That that

Therefore

RN. Therefore

No structures by

were independ by 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

Section cross-reference(s): 76

FAN.CNT 1

=>

L'AIN.	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI PRAI	US 2003194630 US 2001-1756	A 1	20031016 20011024	US 2001-1756	20011024 <

MARPAT 139:314469

Bistable mols. are provided with at least one photosensitive functional group. As thus constituted, the bistable mols. are photo-patternable, thereby allowing fabrication of micrometer-scale and nanometer-scale circuits in discrete areas without relying on a top conductor as a mask. The bistable mols. may comprise mols. that undergo redox reactions, such as rotaxanes and catenanes, or may comprise mols. that undergo an elec.-field-induced band gap change that causes the mols., or a portion thereof, to rotate, bend, twist, or otherwise change from a substantially fully conjugated state to a less conjugated state. The change in states in the latter case results in a change in elec. conductivity

photoresist photomask bistable mol circuitry ST

IT Optical switches

(bistable; photo-patternable mol. circuitry)

Photomasks (lithographic masks) IT

Photoresists

(photo-patternable mol. circuitry)

KATHLEEN FULLER EIC 1700 REMSEN 4B28 571/272-2505

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STRUCTURE FILE UPDATES: 17 MAR 2004 HIGHEST RN 664302-53-8 DICTIONARY FILE UPDATES: 17 MAR 2004 HIGHEST RN 664302-53-8

TSCA INFORMATION NOW CURRENT THROUGH JANUARY 6, 2004

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Experimental and calculated property data are now available. For more information enter HELP PROP at an arrow prompt in the file or refer to the file summary sheet on the web at: http://www.cas.org/ONLINE/DBSS/registryss.html

=> FILE HCAPLU
FILE 'HCAPLUS' ENTERED AT 13:26:20 ON 18 MAR 2004
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FILE COVERS 1907 - 18 Mar 2004 VOL 140 ISS 12 FILE LAST UPDATED: 17 Mar 2004 (20040317/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> D QUE L36 STR

VAR G1=19/25 VAR G2 = 39/1VAR G4=O/S/N/PREP G6=(0-10) A NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM GGCAT IS UNS AT 2 IS PCY UNS AT 30 GGCAT GGCAT IS UNS AT 40 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 38

STEREO ATTRIBUTES: NONE

L38

4 SEA FILE=REGISTRY SSS FUL L36

L39

3 SEA FILE=HCAPLUS ABB=ON L38

=> D L39 ALL 1-3 HITSTR

L39 ANSWER 1 OF 3 HCAPLUS COPYRIGHT 2004 ACS on STN

1991:257120 HCAPLUS AN

114:257120 DN

Entered STN: 28 Jun 1991 ED

Polynuclear azobenzene derivatives as dichroic dyes and liquid-crystal compositions for guest-host displays

Ozawa, Tetsuo; Hosogai, Hisayo IN

Mitsubishi Kasei Corp., Japan PA

Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DTPatent

Japanese LA

ICM C07C245-08

ICS C07C245-10; C07D215-38; C07D215-40; C07D295-12; C07D455-04; C09K019-60; G02F001-13

74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

Section cross-reference(s): 41

FAN.CNT 1

PATENT NO. KIND DATE

APPLICATION NO. DATE

KATHLEEN FULLER EIC 1700 REMSEN 4B28 571/272-2505

query covers Claim 1\$ + 14

19890426 JP 1989-104559 19901126 Α2 JP 02286653 ΡI 19890426 PRAI JP 1989-104559 GΙ

Me (CH₂) 4
$$\sim$$
 C \equiv C \sim NHCH₂ \sim (CH₂) 4Me \sim II

R1AC.tplbond.CA1C.tplbond.CA2N:NA3N:NA4R2 (I; R1 = alkyl, alkoxy; R2 = AΒ NR3R4, NR5CH2C6H4R6-4, 1-pyrrolidinyl, morpholino; R3, R4 = alkyl; R5 = H, alkyl; R6 = H, alkyl, alkoxy; A-A4 = 1,4-phenylene which may be substituted with halo or alkyl, 1,4-naphthylene, 5,8-quinolinediyl; when A4 is substituted with alkyl group, the alkyl group may be bonded with R3, R4, or R5 to form a N-containing alicyclic ring) and liquid-crystal compns. containing I are claimed. I show high dichroic ratio and solubility in liquid crystals and the I-containing liquid-crystal compns. give color displays with fine contrast. A mixture of 4-iodotolan derivative II, CuI, 4-Me(CH2)4C6H4C.tplbond.CH, Et3N, o-C6H4Cl2, Pd(PPh3)2Cl2, and PPh3 was stirred at 50° for 3 h to give I [R1 = pentyl, R2 = NHCH2C6H4 (CH2) 4Me-4, A = A1 = A2 = 1,4-phenylene, A3 = 1,4-naphthylene, A4 = 2-methyl-1,4-phenylene) (III). A composition dissolving III in ZLI-1565 gave a guest-host color display devices.

tolanylethynylphenylazophenylazobenzene dichroic liq crystal display; ST

azobenzene deriv dichroic dye display

Dyes TI

(dichroic, polynuclear tolan-azobenzene derivs., for guest-host liquid-crystal displays)

Optical imaging devices IT

(liquid-crystal, guest-host, polynuclear tolan-azobenzene derivs. as dichroic dyes for)

134240-77-0 IT

RL: USES (Uses)

(condensation of, with (pentylphenyl)acetylene, dichroic dye for quest-host liquid-crystal displays from)

79887-10-8 IT

RL: USES (Uses)

(condensation of, with [[(benzylamino)phenylazo]naphthylazo]iodotolanes dichroic dye for guest-host liquid-crystal displays from)

134216-35-6P 134216-36-7P 134216-37-8P 134216-34-5P IT 134216-33-4P

134216-41-4P 134216-38-9P **134216-39-0P 134216-40-3P**

134216-44-7P 134216-45-8P 134216-46-9P 134216-42-5P 134216-43-6P

134216-51-6P 134216-49-2P 134216-50-5P 134216-47-0P 134216-48-1P

134264-47-4P 134216-54-9P 134216-52-7P 134216-53-8P

RL: PREP (Preparation)

(preparation of, as dichroic dye for guest-host liquid-crystal displays)

IT 134216-39-0P 134216-40-3P RL: PREP (Preparation)

(preparation of, as dichroic dye for guest-host liquid-crystal displays)

RN 134216-39-0 HCAPLUS

CN

Benzenamine, 4-[[4-[[4-[(4-butyl-2-methylphenyl)ethynyl]-1-naphthalenyl]ethynyl]-3-methylphenyl]azo]-2,5-dimethylphenyl]azo]-N-ethyl-N-methyl-(9CI) (CA INDEX NAME)

$$\begin{array}{c} Me \\ N=N \end{array}$$

$$\begin{array}{c} Me \\ N=N \end{array}$$

$$\begin{array}{c} Me \\ N=N \end{array}$$

$$\begin{array}{c} Me \\ Me \end{array}$$

$$\begin{array}{c} N=N \end{array}$$

$$\begin{array}{c} Me \\ Me \end{array}$$

$$\begin{array}{c} N=N \end{array}$$

RN 134216-40-3 HCAPLUS

CN Benzenemethanamine, 4-butyl-N-[4-[[4-[[4-[[4-[[2-fluoro-4-[[4-(2-methylbutoxy)-1-naphthalenyl]ethynyl]phenyl]ethynyl]phenyl]azo]-2,3-dimethylphenyl]azo]-3-methylphenyl]- (9CI) (CA INDEX NAME)

PAGE 1-A

$$\begin{array}{c} \text{Me} \\ \text{N} = \text{N} \\ \text{N} = \text{N} \\ \text{N} = \text{N} \\ \text{Et-CH-CH}_2 = \text{O} \end{array}$$

PAGE 1-B

L39 ANSWER 2 OF 3 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1990:160778 HCAPLUS

DN 112:160778

ED Entered STN: 28 Apr 1990

TI Polarizer plates coated with azo dyes

IN Suzuki, Shinji; Numa, Tatsuya; Danjo, Hideo; Toda, Junji

PA Nippon Kayaku Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G02B005-30 ICS C09B067-22

CC 42-13 (Coatings, Inks, and Related Products)

Section cross-reference(s): 41

FAN.CNT 1

ľA	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	JP 01265205 JP 2568882	A2 B2	19891023 19970108	JP 1988-93379	19880418
PR	AI JP 1988-93379		19880418		
os	MARPAT 112:1607	78			
GΙ					

$$R^{1}$$
 R^{2}
 R^{2}
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 R^{5}
 R^{7}
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 R^{5}
 R^{5

The title coatings providing polarization in any direction, are suitable AΒ for continuous pattern formation and contain I and II and/or III [A1 = (un) substituted phenyl; A2 = Ph, sulfophenyl; R1, R4 = H, OH, lower alkoxy; R2, R3, R5, R6 = H, OH, SO3H; R7 = H, Me, (un) substituted phenyl; R8, R9 = H, Me, MeO, SO3H; B, C = (un)substituted naphthyl; D = (un) substituted Ph, naphthyl; R10 = H, Me, Ac, CONH2, (un) substituted Ph or benzoyl]. A dye solution was prepared from I (A1 = A2 = m-C6H4SO3H; R4 = OEt; R1 = R2 = R6 = R7 = H; R3 = 6'-SO3H; R5 = 7'-SO3H) 2.0, II (B = 4.8-disulfo-2-naphthyl; R8 = OMe; R9 = Me) 1.5, IV 1.5, Emulgen 920 0.05, and 100 parts. An 80 μ -thick cellulose triacetate film was given 10 rubs with a felt on its entire surface in the longitudinal direction and then 20 rubs at a certain interval in the transverse direction, washed with water, dried, coated with the above dye solution, and dried at 60° to give a polarizer plate with 36.0% visible transmittance and 80.0% average polarization.

ST azo dye light polarizer plate

IT Optical imaging devices

(azo dye-coated cellulose triacetate films, light-polarizing)

IT Polarizers

(cellulose triacetate films coated with azo dyes)

IT Dyes, azo

(cellulose triacetate films coated with, for light polarizers)

27990-34-7 IT 15999-06-1 25180-30-7 25738-24-3 25784-17-2 126343-29-1 124521-17-1 124521-18-2 125091-19-2 126343-28-0 126343-34-8 126343-30-4 126343-31-5 126343-32-6 126343-33-7 126343-39-3 126343-35-9 126343-36-0 126343-37-1 126343-38-2 126343-41-7 126343-42-8 126343-43-9 126343-44-0 126343-40-6 126343-45-1 **126343-46-2** 126343-47-3 126343-48-4 126343-49-5 126343-50-8 126343-51-9 126343-52-0 126343-53-1 126343-54-2 126343-55-3 126412-62-2

RL: USES (Uses)

(cellulose triacetate films coated with, light-polarizing)

IT 9012-09-3, Cellulose triacetate

RL: USES (Uses)

(films, azo dye-coated, for light polarizers)

IT 126343-46-2

RL: USES (Uses)

(cellulose triacetate films coated with, light-polarizing)

RN 126343-46-2 HCAPLUS

2-Naphthalenesulfonic acid, 5-[[6-amino-1-hydroxy-3-sulfo-5-[(3-sulfophenyl)azo]-2-naphthalenyl]azo]-8-[[4-[(2,4-dimethyl-6-sulfophenyl)azo]-6-sulfo-1-naphthalenyl]azo]-6-ethoxy- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 2-A

L39 ANSWER 3 OF 3 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1989:9722 HCAPLUS

DN 110:9722

ED Entered STN: 06 Jan 1989

TI Water-based jet-printing ink compositions

IN Ariga, Tamotsu; Hashimoto, Mitsuru; Shimada, Masaru

PA Ricoh Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C09D011-00

ICS C09B067-24; C09D011-00; C09D011-02

CC 41-3 (Dyes, Organic Pigments, Fluorescent Brighteners, and Photographic Sensitizers)

Section cross-reference(s): 42

FAN.CNT 1

- * STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY AVAILABLE VIA OFFLINE PRINT *
- The title ink compns., giving clear images with good water and light resistance, contain ≥1 water-soluble dye I [A = (un)substituted aryl; R1 = OH, (un)substituted amino; M = H, alkali metal, quaternary ammonium, amine; R2 = H, OH, (un)substituted amino; m = 0, 1; n = 1, 2].
 4-O2NC6H4NH2 was diazotized and coupled with 1-amino-8-naphthol-3,6-

disulfonic acid Na salt, and the azo intermediate coupled with diazotized 2,7-diamino-9-fluorenone to give II. II 3.0, diethylene glycol 15.0, N-methyl-2-pyrrolidone 15.0, Na dehydroacetate 0.2, and H2O 66.8% were mixed and stirred at .apprx.50°, and filtered to give an ink. When the ink was applied to jet printing, a clear image was obtained without nozzle clogging. The image showed fading ratio (after 1-min immersion in water at 30°) 8%, and (after 3-h in a fadeometer at 63°) 6%, vs. 7, and 3, resp., for a control ink prepared using C.I. Direct Black 32 instead of II.

jet printing ink azo dye; water resistance jet printing ink; light ST resistance jet printing ink

IT Dyes, azo

(jet printing inks containing, clogging-resistant, giving clear image with good resistance to water and light)

Water-resistant materials TΤ

(inks, jet-printing, containing water-soluble azo dyes, light- and clogging-resistant)

IT

(jet-printing, light- and water-resistant, water-soluble azo dyes for, clogging-resistant)

2915-84-6 IT

RL: USES (Uses)

(coupling of diazotized, with naphthol azo compds.)

100-01-6, p-Nitroaniline, reactions IT

RL: RCT (Reactant); RACT (Reactant or reagent) (coupling of diazotized, with naphthols)

3963-80-2 IT

RL: RCT (Reactant); RACT (Reactant or reagent) (coupling of, with diazotized nitroaniline)

115898-88-9P 115898-90-3P 115898-86-7P 115898-84-5P IT 115898-83-4P 115898-94-7P 115898-95-8P 115898-92-5P **115898-93-6P** 115898-97-0P 115898-98-1P 117869-96-2P 115899-00-8P 115898-96-9P RL: IMF (Industrial manufacture); PREP (Preparation) (preparation of, as dye for jet-printing inks)

115898-93-6P IT

RL: IMF (Industrial manufacture); PREP (Preparation) (preparation of, as dye for jet-printing inks)

115898-93-6 HCAPLUS RN

Benzoic acid, 4,4'-[(9-oxo-9H-fluorene-2,7-diyl)bis[azo(1-amino-8-hydroxy-CN 4,6-disulfo-7,2-naphthalenediyl)azo]]bis[3-methyl-, hexasodium salt (9CI) (CA INDEX NAME)

PAGE 1-A

HO₂C
$$N=N$$
 $N=N$ $N=N$ $N=N$ $N=N$ $N=N$

PAGE 1-B

=> => D QUE L13 L13 1870 SEA FILE=HCAPLUS ABB=ON OPTICAL SWITCHES/IT

=> D QUE L33 L4

Cb \(^{3} \cdot^{A} \cdot^{A} \cdot^{Cb} \cdot^{G2} \cdot^{A} \cdot^{A} \cdot^{Cy} \cdot^{G1} \)
32 31 30 29 28 27 26 1 2 3

STR

A ~ A ~ G5 ~ Cb @19 20 21 22

24 5 10 13 A G6 4 C 6 G4 11 C c 14 @25 C C C C 14 9 C 7 12 C 15 8 16 covera 11,13,14

12 structures

VAR G1=19/25
REP G2=(1-10) A
REP G3=(0-10) A
VAR G4=0/S/N/P
REP G5=(0-10) A
REP G6=(0-10) A
NODE ATTRIBUTES:
DEFAULT MLEVEL IS AT

NODE ATTRIBUTES:
DEFAULT MLEVEL IS ATOM
GGCAT IS UNS AT 2
GGCAT IS UNS AT 22
GGCAT IS UNS AT 28
GGCAT IS UNS AT 32
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 29

STEREO ATTRIBUTES: NONE

DILLIC	TIT TAKE DOT.		. 0.1. 🗀		
L10	400	SEA	FILE=HCAPLUS	ABB=ON	MOLECUL?(3A)CIRCUIT?
L11	4670	SEA	FILE=HCAPLUS	ABB=ON	BISTABIL?
L12	120	SEA	FILE=HCAPLUS	ABB=ON	BISTABIL? (5A) MOLECUL?
L13	1870	SEA	FILE=HCAPLUS	ABB=ON	OPTICAL SWITCHES/IT
L15	37	SEA	FILE=HCAPLUS	ABB=ON	L11 AND L13
L16	220	SEA	FILE=HCAPLUS	ABB=ON	L13 AND MOLEC?

```
764 SEA FILE=HCAPLUS ABB=ON L12 OR L10 OR L15 OR L16
L17
             9 SEA FILE=HCAPLUS ABB=ON L11 AND PHOTOSENSIT?
L18
           101 SEA FILE=HCAPLUS ABB=ON L11 AND CIRCUIT?
L19
                                       (L17 OR L18 OR L19)
           868 SEA FILE=HCAPLUS ABB=ON
L20
          2798 SEA FILE=HCAPLUS ABB=ON MOLECUL? (5A) SWITCH?
L22
          3504 SEA FILE=HCAPLUS ABB=ON L20 OR L22
L23
               SEL L23 1- RN:
                                   8549 TERMS
L24
          8549 SEA FILE=REGISTRY ABB=ON L24
L25
           149 SEA FILE=HCAPLUS ABB=ON L11 AND ?RESIST?
L27
               SEL L27 1- RN:
                                    167 TERMS
L28
           167 SEA FILE=REGISTRY ABB=ON L28
L29
           8639 SEA FILE=REGISTRY ABB=ON L25 OR L29
L30
           _12 SEA FILE=REGISTRY SUB=L30 SSS FUL L4
L32
                                            13 CA references
            13 SEA FILE=HCAPLUS ABB=ON L32
L33
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=> D L33 ALL 1-13 HITSTR

L33 ANSWER 1 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

2003:139800 HCAPLUS ΑN

138:321666 DN

Entered STN: 25 Feb 2003 ED

- Single Molecule Force Spectroscopy of Azobenzene Polymers: Switching TΤ Elasticity of Single Photochromic Macromolecules
- Holland, Nolan B.; Hugel, Thorsten; Neuert, Gregor; Cattani-Scholz, Anna; AU Renner, Christian; Oesterhelt, Dieter; Moroder, Luis; Seitz, Markus; Gaub, Hermann E.
- Lehrstuhl fuer Angewandte Physik Center for Nanoscience, CS Ludwig-Maximilians-Universitaet, Munich, 80799, Germany
- Macromolecules (2003), 36(6), 2015-2023 SO CODEN: MAMOBX; ISSN: 0024-9297
- American Chemical Society PB
- DTJournal
- English T.A
- 35-5 (Chemistry of Synthetic High Polymers) CC Section cross-reference(s): 22, 34, 73
- The reversible, optical switching of individual mols. of a polypeptide AΒ with azobenzene moieties, was observed using mol. force spectroscopy. The polypeptide was prepared by polycondensation of tripeptide monomers containing (4-aminomethyl)phenylazobenzoic acid (AMPB) to obtain H-Cys(Trt)-[Lys(Adoc)-AMPB-Gly]n-OH. The contour length of the polymer could be selectively lengthened or shortened by switching between the trans- and cis-azo configurations with 420 and 365 nm wavelength light, resp. This cis- to trans-azo configurational transition induced by UV light resulted in a measurable change in polymer contour length. The contour length change was observed at low force and under external loads of up to 400 pN using a modified force spectrometer, in which the sample could be irradiated in total internal reflectance. The ability to shorten the polymer against an external load demonstrates photomech. energy conversion in an individual mol., of interest in development of mol. machines.
- polypeptide aminomethylphenylazobenzoic acid moiety reversible switching; single mol photochromic polypeptide azobenzene chromophore elasticity switching; photomech energy conversion elasticity switching polypeptide azobenzene
- $_{
 m IT}$ Plasticity

(photoplasticity; switching elasticity of polypeptide-azobenzene single photochromic mols. studied by optical excitation and AFM using object slide as waveguide)

IT Photomechanical effect

> (plasticity; switching elasticity of polypeptide-azobenzene single photochromic mols. studied by optical excitation and AFM using object slide as waveguide)

IT Atomic force microscopy

Chromophores

Molecular dynamics

Optical switching

Optical waveguides

Photochromism

Photoelasticity

(switching elasticity of polypeptide-azobenzene single photochromic mols. studied by optical excitation and AFM using object slide as waveguide)

IT Polyamides, properties

RL: PRP (Properties)

(switching elasticity of polypeptide-azobenzene single photochromic mols. studied by optical excitation and AFM using object slide as waveguide)

103213-32-7DP, termination products with peptides IT 512197-42-1DP, cysteine derivative-terminated 512197-45-4P

RL: PNU (Preparation, unclassified); PRP (Properties); PREP (Preparation) (switching elasticity of polypeptide-azobenzene single photochromic mols. studied by optical excitation and AFM using object slide as waveguide)

THERE ARE 72 CITED REFERENCES AVAILABLE FOR THIS RECORD RE.CNT RE

- (1) Anon; CRC Press 1989, V2, P119
- (2) Archut, A; J Am Chem Soc 1998, V120, P12187 HCAPLUS
- (3) Arrhenius, S; Z Phys Chem 1889, V4, P226
- (4) Asanuma, H; Angew Chem, Int Ed 1999, V38, P2393 HCAPLUS
- (5) Ashkin, A; Nature (London) 1990, V348, P346 MEDLINE
- (6) Ballardini, R; Acc Chem Res 2001, V34, P445 HCAPLUS
- (7) Balzani, V; Angew Chem, Int Ed 2000, V39, P3348 HCAPLUS
- (8) Balzani, V; Proc Natl Acad Sci U S A 2002, V99, P4814 HCAPLUS
- (9) Behrendt, R; Angew Chem, Int Ed 1999, V38, P2771 HCAPLUS
- (10) Behrendt, R; J Pept Sci 1999, V5, P519 HCAPLUS
- (11) Bell, G; Science 1978, V200, P618 HCAPLUS
- (12) Bieringer, T; Macromol Chem Phys 1995, V196, P1375 HCAPLUS
- (13) Binnig, G; Phys Rev Lett 1986, V56, P930
- (14) Blair, H; Polymer 1980, V21, P1195 HCAPLUS
- (15) Bouchiat, C; Biophys J 1999, V76, P409 HCAPLUS
- (16) Brown, C; Photochromism 1971
- (17) Bustamante, C; Acc Chem Res 2001, V34, P412 HCAPLUS
- (18) Bustamante, C; Science 1994, V265, P1599 MEDLINE
- (19) Butt, H; Nanotechnology 1995, V6, P1
- (20) Clausen-Schaumann, H; Curr Opin Chem Biol 2000, V4, P524 HCAPLUS
- (21) Collin, J; Acc Chem Res 2001, V34, P477 HCAPLUS
- (22) Eich, M; Makromol Chem Rapid Commun 1987, V8, P467 HCAPLUS
- (23) Eich, M; Makromol Chem Rapid Commun 1987, V8, P59 HCAPLUS
- (24) Eisenbach, C; Polymer 1980, V21, P1175 HCAPLUS (25) Feringa, B; Chem Rev 2000, V100, P1789 HCAPLUS
- (26) Feringa, B; Tetrahedron 1993, V49, P8267 HCAPLUS
- (27) Finkelmann, H; Phys Rev Lett 2001, V87, P015501 MEDLINE
- (28) Grandbois, M; Science 1999, V283, P1727 HCAPLUS
- (29) Hanggi, P; Rev Mod Phys 1990, V62, P251
- (30) Hartley, G; Nature (London) 1937, V140, P281 HCAPLUS
- (31) Howard, J; Mechanics of Motor Proteins and the Cytoskeleton 2001

```
(32) Hugel, T; Macromol Rapid Commun 2001, V22, P989 HCAPLUS
(33) Hugel, T; Science 2002, V296, P1103
(34) Ikeda, T; Science 1995, V268, P1873 HCAPLUS
(35) Irie, M; Adv Polym Sci 1990, V94, P27 HCAPLUS
(36) Irie, M; Macromolecules 1981, V14, P262 HCAPLUS
(37) Irie, M; Photoreactive Materials for Ultrahigh-Density Optical Memory 1994
(38) Janshoff, A; Angew Chem, Int Ed 2000, V39, P3212 HCAPLUS
(39) Keller, D; Biophys J 2000, V78, P541 HCAPLUS
(40) Kishino, A; Nature (London) 1988, V334, P74 MEDLINE
(41) Kumar, G; Chem Rev 1989, V89, P1915 HCAPLUS
(42) Labarthet, F; Phys Chem Chem Phys 2000, V2, P5154 HCAPLUS
(43) Livadaru, L; Submitted to Macromolecules
(44) Monti, S; Chem Phys 1982, V71, P87 HCAPLUS
(45) Nagele, T; Chem Phys Lett 1997, V272, P489
(46) Oesterhelt, F; New J Phys 1999, V1, P6.1
(47) Oster, G; ATP Synthase: The rotary molecular motors working together 1999
(48) Pease, A; Acc Chem Res 2001, V34, P433 HCAPLUS
(49) Rau, H; J Photochem 1984, V26, P221 HCAPLUS
(50) Rau, H; Photochromism: Molecules and Systems 1990, V40, P165 HCAPLUS
(51) Renner, C; Biopolymers 2000, V54, P501 HCAPLUS
(52) Renner, C; Biopolymers 2002, V63, P382 HCAPLUS
(53) Rief, M; Nat Struct Biol 1999, V6, P346 HCAPLUS
(54) Rief, M; Phys Rev Lett 1998, V81, P4764 HCAPLUS
(55) Rief, M; Science 1997, V275, P1295 HCAPLUS
(56) Robertson, J; J Chem Soc 1939, P232 HCAPLUS
(57) Schalley, C; Acc Chem Res 2001, V34, P465 HCAPLUS
(58) Schulz, B; Synth Met 2001, V124, P155 HCAPLUS
(59) Sheetz, M; Laser Tweezers in Cell Biology 1997
(60) Shinkai, S; Top Curr Chem 1984, V121, P67 HCAPLUS
(61) Shipway, A; Acc Chem Res 2001, V34, P421 HCAPLUS
(62) Smith, S; Science 1992, V258, P1122 HCAPLUS (63) Smith, S; Science 1996, V271, P795 HCAPLUS
(64) Strzegowski, L; J Am Chem Soc 1994, V116, P813 HCAPLUS
(65) Tamai, N; Chem Rev 2000, V100, P1875 HCAPLUS
(66) Ulysse, L; J Am Chem Soc 1995, V117, P8466 HCAPLUS
(67) Vale, R; Science 2000, V288, P88 HCAPLUS
(68) Viani, M; Rev Sci Instrum 1999, V70, P4300 HCAPLUS
(69) Vogtle, F; Supramolecular Chemistry 1991
(70) Willner, I; Acc Chem Res 1997, V30, P347 HCAPLUS
(71) Wurthner, F; J Chem Soc, Perkin Trans 2 1995, P1727
(72) Zilker, S; Adv Mater 1998, V10, P855 HCAPLUS
IT
         512197-45-4P
         RL: PNU (Preparation, unclassified); PRP (Properties); PREP (Preparation)
               (switching elasticity of polypeptide-azobenzene single photochromic
              mols. studied by optical excitation and AFM using object slide as
              waveguide)
         512197-45-4 HCAPLUS
RN
         Poly[(1E)-azo-1,4-phenylenecarbonylimino(2-oxo-1,2-ethanediyl)imino[(1S)-2-
CN
         oxo-1-[4-[[(tricyclo[3.3.1.13,7]dec-1-yloxy)carbonyl]amino]butyl]-1,2-inverse and the second control of the 
         ethanediyl]iminomethylene-1,4-phenylene], \alpha-[4-[[[S-
         (triphenylmethyl)-L-cysteinyl-N6-[(tricyclo[3.3.1.13,7]dec-1-
         yloxy) carbonyl]-L-lysyl] amino] methyl] phenyl] -\omega-[(1E)-[4-
         [[(carboxymethyl)amino]carbonyl]phenyl]azo]- (9CI) (CA INDEX NAME)
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PAGE 1-A

PAGE 2-B

L33 ANSWER 2 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2001:330311 HCAPLUS

DN 135:84571

ED Entered STN: 09 May 2001

KATHLEEN FULLER EIC 1700 REMSEN 4B28 571/272-2505

```
Liquid crystalline dimeric compounds with an alkylene spacer
ΤI
     Prasad, Veena; Rao, D. S. Shankar; Prasad, S. Krishna
ΑU
     Centre for Liquid Crystal Research, Bangalore, 560 013, India
CS
     Liquid Crystals (2001), 28(5), 761-767
SO
     CODEN: LICRE6; ISSN: 0267-8292
     Taylor & Francis Ltd.
PΒ
DT
     Journal
LΑ
     English
     75-11 (Crystallography and Liquid Crystals)
CC
     Section cross-reference(s): 25, 73
     Two series of compds. with dimeric mols. were synthesized and
AΒ
     characterized. These mols. consist of two nonmesogenic units linked by an
     alkylene spacer through an ester linkage. The effects of varying the
     terminal as well as the spacer chain length on the mesomorphic properties
     were studied. For compds. with an even number of C atoms in the spacer
     chains, the observed textures in the mesophases are reminiscent of those seen
     for the smectic phases of banana-shaped mols.
     liq crystal dimeric compd alkylene space ester linkage
ST
IT
     Phase transition enthalpy
        (of liquid crystal dimeric compds. with alkylene spacer through ester
        linkage)
IT
     Liquid crystals
        (preparation and properties of dimeric compds. with alkylene spacer through
        ester linkage)
     Electrooptical effect
IT
        (switching; of liquid crystal dimeric compound with alkylene spacer through
        ester linkage)
IT
     Liquid crystals
        (transitions; of dimeric compds. with alkylene spacer through ester
        linkage)
     280116-08-7
IT
                   294624-56-9
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (esterification of)
                                    347151-00-2P
                                                   347151-02-4P
IT
     280116-07-6P
                    347150-98-5P
                                                   347151-10-4P
                                    347151-08-0P
                                                                  347151-12-6P
     347151-04-6P
                    347151-06-8P
                                                   347151-20-6P
     347151-14-8P
                    347151-16-0P
                                    347151-18-2P
                                                                  347151-22-8P
     347151-24-0P
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN
     (Synthetic preparation); PREP (Preparation); PROC (Process)
        (preparation and liquid crystal properties of)
              THERE ARE 20 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT
(1) Blumstein, R; Mol Cryst liq Cryst 1985, V129, P375 HCAPLUS
(2) Choi, S; Mol Cryst liq Cryst 1999, V328, P185 HCAPLUS
(3) Emsley, J; Mol Cryst liq Cryst Lett 1984, V102, P223 HCAPLUS
(4) Griffin, A; J Am chem Soc 1981, V103, P4957 HCAPLUS
(5) Hassner, A; Tetrahedron Lett 1978, P4475 HCAPLUS
(6) Imrie, C; Handbook of Liquid Crystals 1998, V2B HCAPLUS
(7) Jin, J; Bull Korean chem Soc 1985, V6, P40
(8) Jin, J; J chem Soc, Perkin Trans II 1986, P343 HCAPLUS
(9) Jin, J; Polym J (Jpn) 1986, V18, P63 HCAPLUS
(10) Pelzl, G; Liq Cryst 1999, V26, P135 HCAPLUS
(11) Pelzl, G; Liq Cryst 1999, V26, P401 HCAPLUS
(12) Pociecha, D; Mol Cryst liq Cryst 1999, V328, P75 HCAPLUS
(13) Prasad, V; Liq Cryst 2000, V27, P585 HCAPLUS
(14) Prasad, V; Liq Cryst (communicated)
(15) Sekine, T; Jpn J appl Phys 1997, V36, P6455 HCAPLUS
(16) Shen, D; Chem Commun 1998, P2573 HCAPLUS
```

(17) Shilov, S; Liq Cryst 1999, V9, P1409

(18) Vorlander, D; Z phys Chem 1927, V126, P449 HCAPLUS

(19) Weissflog, W; Mol Cryst liq Cryst 1999, V328, P101 HCAPLUS

(20) Weissflog, W; Proc SPIE 1998, V3319, P14 HCAPLUS

RN 280116-07-6 HCAPLUS

CN Hexanedioic acid, bis[4-[(E)-[[4-(decyloxy)phenyl]imino]methyl]phenyl] ester (9CI) (CA INDEX NAME)

Double bond geometry as shown.

PAGE 1-B

- L33 ANSWER 3 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN
- AN 2000:764376 HCAPLUS
- DN 134:49469
- ED Entered STN: 01 Nov 2000
- TI Smectic mesophase properties of dimeric compounds. 2. Distinct formation of smectic structures with antiferroelectric ordering and frustration
- AU Watanabe, Junji; Izumi, Tatuya; Niori, Teruki; Zennyoji, Masahito; Takanishi, Yoichi; Takezoe, Hideo
- CS Department of Polymer Chemistry, Tokyo Institute of Technology, Tokyo, 152, Japan
- Molecular Crystals and Liquid Crystals Science and Technology, Section A:
 Molecular Crystals and Liquid Crystals (2000), 346, 77-86
 CODEN: MCLCE9; ISSN: 1058-725X
- PB Gordon & Breach Science Publishers
- DT Journal
- LA English
- CC 75-11 (Crystallography and Liquid Crystals) Section cross-reference(s): 25, 76

Two series of dimeric compds., which comprise the Schiff's base mesogen, AΒ alkyl spacer with the C number of 5 and alkyl or alkoxy tail with the C number of m = 4-16, were prepared In these compds., three types of smectic liquid crystals were observed: the single layer phase with the tail group randomly mixing with the spacer, the bilayer phase in which the segregation of the spacer and tail groups takes place and so two mesogenic layers are included within a repeat unit, and the frustrated smectic phase in which the d. modulation appears along the layer as well as the layer normal. The bilayer phase is antiferroelec. and the frustrated smectic phase was considered to result from the two-dimensional escape from the dipolar interaction. The phase behavior with m and the structure and properties of each phase are described. dimeric compd Schiff base mesogenic group smectic antiferroelec ST Liquid crystals IT(antiferroelec.; preparation and phase behavior and structure of dimeric compds. with Schiff base mesogenic group) Antiferroelectric materials IT(liquid-crystal; preparation and phase behavior and structure of dimeric compds. with Schiff base mesogenic group) IT Phase transition enthalpy Phase transition entropy (of dimeric compds. with Schiff base mesogenic group) ITLiquid crystals (smectic; preparation and phase behavior and structure of dimeric compds. with Schiff base mesogenic group) IT Liquid crystals (transitions; of dimeric compds. with Schiff base mesogenic group) 211049-26-2P **243975-13-5P 250689-35-1P** IT 312610-29-0P 312610-33-6P 312610-34-7P 312610-30-3P 312610-31-4P RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process) (preparation and liquid crystal phase behavior and structure of) IT 312610-32-5P RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process) (preparation and thermal behavior of) THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD RE.CNT (1) Attard, G; Liq Cryst 1994, V16, P529 HCAPLUS(2) Blatch, A; Liq Cryst 1995, V18, P801 HCAPLUS (3) Date, R; Liq Cryst 1992, V12, P203 HCAPLUS (4) Nakata, Y; J Phys II (France) 1994, V4(581), P581 (5) Nakata, Y; Polym J 1997, V29, P193 HCAPLUS (6) Niori, T; J Mater Chem 1996, V6, P1231 HCAPLUS (7) Niori, T; Liq Cryst 1995, V19, P139 HCAPLUS (8) Takanishi, Y; Phys Rev B 1992, V45, P7684 HCAPLUS (9) Takanishi, Y; to be published in Jpn J Appl Phys (10) Watanabe, J; Jpn J Appl Phys 1998, V37, PL401 HCAPLUS (11) Watanabe, J; Liq Cryst 1993, V13, P455 HCAPLUS (12) Watanabe, J; Prog Polym Sci 1997, V22, P1053 HCAPLUS 243975-13-5P 250689-35-1P RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process) (preparation and liquid crystal phase behavior and structure of) RN 243975-13-5 HCAPLUS

Benzoic acid, 4-[(E)-[(4-dodecylphenyl)imino]methyl]-, 1,5-pentanediyl

ester (9CI) (CA INDEX NAME)

CN

CHU 10/001756 3/17/04 Page 18

Double bond geometry as shown.

PAGE 1-B

RN 250689-35-1 HCAPLUS
CN Benzoic acid, 4-[(E)-[[4-(octyloxy)phenyl]imino]methyl]-, 1,5-pentanediyl ester (9CI) (CA INDEX NAME)

Double bond geometry as shown.

PAGE 1-B

L33 ANSWER 4 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2000:359759 HCAPLUS

DN 133:81805

ED Entered STN: 31 May 2000

TI Ferroelectric switching in a novel bent-shaped mesogen having two non-mesogenic units linked by an alkylene spacer

AU Prasad, Veena; Rao, D. S. Shankar; Prasad, S. Krishna

CS Centre for Liquid Crystal Research, Bangalore, 560 013, India

SO Liquid Crystals (2000), 27(5), 585-590 CODEN: LICRE6; ISSN: 0267-8292

PB Taylor & Francis Ltd.

DT Journal

LA English

ST

CC 75-11 (Crystallography and Liquid Crystals) Section cross-reference(s): 25, 74, 76

The authors report the synthesis, x-ray diffraction results and electrooptical switching measurements for a bent-shaped mesogen having two nonmesogenic units linked by an alkylene spacer. The novelty of the mol. structure lies in the carbonyl group of the ester linkage being directly attached to the spacer unit, unlike for banana-shaped mols. reported so far, in which it is one O atom away from the spacer or the central aromatic unit. The compound shows two mesophases: the high temperature mesophase is a tilted smectic phase showing ferroelec. switching characteristics; the low temperature phase is more highly ordered with textural features similar to that of the B3 banana phase.

ferroelec switching bent shaped mesogen nonmesogenic alkylene spacer

IT Ferroelectric switching

(ferroelec. switching in bent-shaped mesogen having two non-mesogenic units linked by alkylene spacer)

IT Molecular structure-property relationship

(ferroelec. switching; in bent-shaped mesogen having two non-mesogenic units linked by alkylene spacer)

IT Liquid crystals

(in bent-shaped mesogen having two non-mesogenic units linked by alkylene spacer)

IT Liquid crystals

(transitions; of bis[(decyloxyphenyliminomethyl)phenyl] adipate)

IT 124-04-9, Hexanedioic acid, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(esterification of adipic acid with (decyloxyphenyliminomethyl)phenol)

IT 280116-08-7P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT

(Reactant or reagent)

(preparation and esterification of adipic acid by)

IT 280116-07-6P

RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)

(preparation, liquid crystal properties and ferroelec. switching of)

IT 123-08-0, 4-Hydroxybenzaldehyde 39905-47-0, 4-(Decyloxy)aniline RL: RCT (Reactant); RACT (Reactant or reagent)

(reaction of decyloxyaniline with hydroxybenzaldehyde)

RE.CNT 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD

- (1) Brand, H; Eur Phys J 1998, VB6, P347
- (2) Choi, S; Mol Cryst liq Cryst 1999, V328, P185 HCAPLUS
- (3) Jakli, A; Phys Rev E 1998, V57, P6737 HCAPLUS
- (4) Kumar, S; J mater Chem 1999, V9, P2751 HCAPLUS
- (5) Link, D; Science 1997, V278, P1924 HCAPLUS
- (6) Niori, T; J mater Chem 1996, V6, P1231 HCAPLUS
- (7) Pelzl, G; Liq Cryst 1999, V26, P401 HCAPLUS
- (8) Sekine, T; Jpn J appl Phys 1997, V36, P6455 HCAPLUS
- (9) Shen, D; Chem Commun 1998, P2573 HCAPLUS
- (10) Shen, D; J mater Chem 1999, V9, P661 HCAPLUS
- (11) Watanabe, J; Jpn J appl Phys 1998, V37, PL401 HCAPLUS
- IT 280116-07-6P

RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)

(preparation, liquid crystal properties and ferroelec. switching of)

RN 280116-07-6 HCAPLUS

CN Hexanedioic acid, bis[4-[(E)-[[4-(decyloxy)phenyl]imino]methyl]phenyl]
 ester (9CI) (CA INDEX NAME)

Double bond geometry as shown.

ANSWER 5 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN AN1999:734438 HCAPLUS DN 132:71699 Entered STN: 19 Nov 1999 EDField-induced molecular reorientation keeping a frustrated structure in an achiral bent-shaped liquid crystal ΑU Takanishi, Yoichi; Izumi, Tatsuya; Watanabe, Junji; Ishikawa, Ken; Takezoe, Hideo; Iida, Atsuo CS Dep. of Organic and Polymeric Materials, Tokyo Institute of Technology, Tokyo, 152-8552, Japan SO Journal of Materials Chemistry (1999), 9(11), 2771-2774 CODEN: JMACEP; ISSN: 0959-9428 PB Royal Society of Chemistry DTJournal English LA CC 75-11 (Crystallography and Liquid Crystals) Section cross-reference(s): 25, 73, 76 The layer and mol. orientational structures were studied by x-ray AΒ microbeam diffraction and optical birefringence measurements in the frustrated smectic phase of a bent-shaped mol. with two mesogens linked by an alkylene spacer $(\alpha, \omega-bis\{4-[(4$ octyloxyphenyl)iminomethyl]benzoyloxy}pentane). In the x-ray microbeam measurement, only one diffraction peak indicating (002) was observed in a thin homogeneous cell without an elec. field, while two other peaks corresponding to (101) and (10.hivin.1) also appear by applying the field. Also, also the birefringence under an applied field is larger than that without the field. Mols. reorient due to dielec. anisotropy keeping the frustrated structure. SŤ octyloxyphenyliminomethylbenzoyloxypentane smectic mol reorientation elec field ΙT Dielectric anisotropy (elec. field-induced mol. reorientation keeping frustrated structure in achiral bent-shaped liquid crystal bis{[(octyloxyphenyl)iminomethyl]benzo yloxy)pentane in relation to) IT Molecular reorientation (field-induced mol. reorientation keeping frustrated structure in achiral bent-shaped liquid crystal bis{[(octyloxyphenyl)iminomethyl]benzo yloxy}pentane) IT Electric field effects (induced mol. reorientation keeping frustrated structure in achiral bent-shaped liquid crystal bis{[(octyloxyphenyl)iminomethyl]benzoyloxy}pe ntane) ΙT Kerr effect (electrooptical) (of achiral bent-shaped liquid crystal bis{[(octyloxyphenyl)iminomethyl]b enzoyloxy}pentane) ΙT Birefringence (of achiral bent-shaped liquid crystal bis{[(octyloxyphenyl)iminomethyl]b enzoyloxy)pentane with and without elec. field) ITLiquid crystals (smectic; field-induced mol. reorientation keeping frustrated structure in achiral bent-shaped liquid crystal bis{[(octyloxyphenyl)iminomethyl]be nzoyloxy}pentane) $\mathbf{T}\mathbf{I}$ 250689-35-1 RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (field-induced mol. reorientation keeping frustrated structure in achiral bent-shaped liquid crystal)

THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE.CNT 11

(1) Chan, K; Phys Rev A 1986, V34, P1420 HCAPLUS

(2) Choi, S; to be published in Mol Cryst Liq Cryst, Proceedings of ILCC98 1999

(3) Fukuda, A; J Mater Chem 1994, V4, P997 HCAPLUS

(4) Iida, A; Ferroelectrics 1994, V149, P117

(5) Itoh, K; J Mater Chem 1997, V7, P407 HCAPLUS

(6) Link, D; Science 1997, V278, P1924 HCAPLUS

(7) Niori, T; J Mater Chem 1996, V6, P1231 HCAPLUS

(8) Niori, T; PhD Thesis, Tokyo Institute of Technology 1997

(9) Sigaud, G; J Phys (France) 1981, V42, P107 HCAPLUS

(10) Takanishi, Y; Jpn J Appl Phys 1996, V35, P683 HCAPLUS

(11) Watanabe, J; Jpn J Appl Phys 1998, V37, PL139 HCAPLUS

250689-35-1 IT

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(field-induced mol. reorientation keeping frustrated structure in achiral bent-shaped liquid crystal)

250689-35-1 HCAPLUS RN

Benzoic acid, 4-[(E)-[[4-(octyloxy)phenyl]imino]methyl]-, 1,5-pentanediyl CN ester (9CI) (CA INDEX NAME)

Double bond geometry as shown.

PAGE 1-B

L33 ANSWER 6 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

1999:649499 HCAPLUS AN

131:345712 DN

Entered STN: 13 Oct 1999 ED

An Iron-Based Molecular Redox Switch as a Model for Iron Release from TI Enterobactin via the Salicylate Binding Mode

```
Ward, Thomas R.; Lutz, Andreas; Parel, Serge P.; Ensling, Juergen;
ΑU
     Guetlich, Philipp; Buglyo, Peter; Orvig, Chris
     Department of Chemistry and Biochemistry, University of Berne, Bern,
CŚ
     CH-3012, Switz.
     Inorganic Chemistry (1999), 38(22), 5007-5017
SO
     CODEN: INOCAJ; ISSN: 0020-1669
     American Chemical Society
PΒ
     Journal
DT
     English
LA
     78-7 (Inorganic Chemicals and Reactions)
CC
     Section cross-reference(s): 6, 65, 68, 72, 73
     The Fe release mechanism from protonated ferric enterobactin
AΒ
     [FeIII(enterobactinH3)] via the salicylate binding mode was probed. For
     this purpose, a tripodal dodecadentate ligand incorporating three
     salicylamide (00) and three bipyridine (NN) binding sites was synthesized
     as well as Fe complexes thereof. A ferric ion coordinates selectively to
     the hard salicylamides and a ferrous ion binds to the softer bipyridines.
     Upon reduction or oxidation, the Fe translocates reversibly and
intramolecularly
     from one site to the other, thus displaying switchlike properties.
     states were characterized by cyclic voltammetry and visible and Mossbauer
     spectroscopy. The Mossbauer spectrum for the ferric complex is fully
     consistent with that obtained by Pecoraro et al. upon lowering the pH of
     [FeIII(enterobactin)]3- solns. (Pecoraro, V. L., et al., 1983), thus
     supporting the alternative Fe release mechanism from enterobactin via the
     salicylate binding mode.
     enterobactin iron release model mol redox switch; iron salicylamide
ST
     bipyridine tripodal ligand prepn redox switch; stability const iron
     salicylamide bipyridine tripodal ligand; protonation const bipyridine
     tripodal ligand; Mossbauer iron salicylamide bipyridine tripodal complex;
     electrochem redn iron salicylamide bipyridine tripodal complex
TT
     Isomers
        (linkage; of iron(II) vs. iron(III) to salicylamide bipyridine tripodal
        ligands)
IT
     Formation constant
     Reduction potential
        (of iron salicylamide bipyridine tripodal complexes)
IT
     Coordination (structure)
        (of iron(II) vs. iron(III) to salicylamide bipyridine tripodal ligands)
IT
     Protonation
        (of salicylamide bipyridine tripodal ligands)
IT
     Molecular structure
        (optimized; of iron salicylamide bipyridine tripodal ligand complexes)
IT
     182870-02-6
                  182870-04-8
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (complexation with iron)
     28384-96-5, Enterobactin
IT
     RL: PEP (Physical, engineering or chemical process); RCT (Reactant); PROC
     (Process); RACT (Reactant or reagent)
        (iron-based mol. redox switch as a model for iron release from
        enterobactin via salicylate binding mode)
IT
     250207-71-7P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and complexation with iron)
IT
     250207-72-8P
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
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(preparation and complexation with iron and protonation consts.)
IT
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation and cyclic voltammetry and optimized geometry)
     250207-73-9P
TT
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation and cyclic voltammetry and optimized geometry and reduction and
        Mossbauer spectrum)
     250207-78-4P
IT
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation and cyclic voltammetry and oxidation with change in chelation
mode
        of ligand)
     250207-74-0P
IT
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation and cyclic voltammetry and stability consts.)
ΙT
     250207-76-2P
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation and oxidation and cyclic voltammetry and Mossbauer spectrum)
     98-88-4, Benzoyl chloride
                                 182870-00-4
TT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reactant for preparation of salicylamide bipyridine tripodal ligands)
     7440-50-8D, Copper, complexes with salicylamide bipyridine tripodal
ΙT
                           182870-04-8D, copper complexes 250207-72-8D
     ligands, properties
      copper complexes
     RL: FMU (Formation, unclassified); PRP (Properties); FORM (Formation,
     nonpreparative)
        (stability consts.)
              THERE ARE 50 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT
RE
(1) Baes, C; Hydrolysis of Cations 1976
(2) Barlow, S; Chem Soc 1996, V118, P7578 HCAPLUS
(3) Belle, C; New J Chem 1998, P1399 HCAPLUS
(4) Blanc, S; J Am Chem Soc 1997, V119, P4934 HCAPLUS
(5) Bonardi, A; J Chem Soc Dalton Trans 1991, P1063 HCAPLUS
(6) Boulas, P; Angew Chem Int Ed Engl 1998, V37, P216 HCAPLUS
(7) Brickman, T; J Biol Chem 1992, V267, P12350 HCAPLUS
(8) Canevet, C; Angew Chem Int Ed Engl 1996, V35, P2657 HCAPLUS
(9) Caravan, P; Inorg Chem 1997, V36, P236 HCAPLUS
(10) Caravan, P; J Am Chem Soc 1995, V117, P11230 HCAPLUS
(11) Cardenas, D; J Am Chem Soc 1996, V118, P11980 HCAPLUS
(12) Cass, M; J Am Chem Soc 1989, V111, P1677 HCAPLUS
(13) Cohen, S; J Am Chem Soc 1998, V120, P6277 HCAPLUS
(14) Constable, E; Nature 1995, V374, P760 HCAPLUS
(15) Credi, A; J Am Chem Soc 1997, V119, P2679 HCAPLUS
(16) Crumbliss, A; Pure Appl Chem 1996, V68, P1225 HCAPLUS
(17) De Santis, G; Inorg Chem 1997, V36, P827
(18) Decurtis, S; J Am Chem Soc 1994, V116, P9521
(19) Gans, P; J Chem Soc, Dalton Trans 1985, P1195 HCAPLUS
(20) Geraldes, C; J Chem Soc Dalton Trans 1992, P2517 HCAPLUS
(21) Gran, G; Acta Chem Scand 1950, V4, P559 HCAPLUS
(22) Greenwood, N; Mossbauer Spectroscopy 1971
(23) Harris, W; J Am Chem Soc 1979, V101, P6097 HCAPLUS
```

- (24) Harris, W; J Am Chem Soc 1981, V103, P2667 HCAPLUS
- (25) Heidinger, S; J Bacteriol 1983, V153, P109 HCAPLUS
- (26) Hider, R; Struct Bonding 1984, V58, P25 HCAPLUS
- (27) James, B; Biochemistry 1962, V1, P379 HCAPLUS
- (28) James, B; J Chem Soc 1961, P4630 HCAPLUS
- (29) Lee, C; J Am Chem Soc 1985, V107, P6920 HCAPLUS
- (30) Lehn, J; Supramolecular Chemistry; Concepts and Perspectives 1995
- (31) Lippard, S; Principles of Bioinorganic Chemistry 1994
- (32) Livoreil, A; J Am Chem Soc 1994, V116, P9399 HCAPLUS (33) Loomis, L; Inorg Chem 1991, V30, P906 HCAPLUS
- (34) Lutz, A; Helv Chim Acta 1998, V81, P207 HCAPLUS
- (35) Lutz, A; Tetrahedron 1996, V52, P12197 HCAPLUS
- (36) Matzanke, B; Iron Carriers and Iron Proteins 1989, P1 HCAPLUS
- (37) Muller, W; MOSFUN Mossbauer Spectra Fitting Program for Universal Theories, Description and User's Guide 1980
- (38) O'Brien, J; Biochim Biophys Acta 1971, V237, P537
- (39) Pecoraro, V; J Am Chem Soc 1983, V105, P4617 HCAPLUS
- (40) Pecoraro, V; J Am Chem Soc 1983, V105, P4623 HCAPLUS
- (41) Petti, G; IUPAC Stability Constants Database 1993
- (42) Raymond, K; Pure Appl Chem 1987, V59, P771 HCAPLUS
- (43) Shanzer, A; J Am Chem Soc 1986, V108, P7609 HCAPLUS
- (44) Sigel, A; Metal Ions Biological Systems 1998, V35
- (45) Stassinopoulos, A; J Am Chem Soc 1991, V113, P8686 HCAPLUS
- (46) Theil, E; Bioinorganic Chemistry 1994, P1 MEDLINE
- (47) Tsubata, Y; J Org Chem 1992, V57, P6749 HCAPLUS
- (48) Zahn, S; Angew Chem Int Ed Engl 1998, V37, P305 HCAPLUS
- (49) Zekany, L; Computational Methods for the Determination of Stability Constants 1985
- (50) Zelikovich, L; Nature 1995, V374, P790 HCAPLUS
- IT 250207-71-7P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and complexation with iron)

- 250207-71-7 HCAPLUS RN
- Benzamide, 4,4',4''-[nitrilotris[2,1-ethanediyl(methylimino)methylene]]tri CN s[N-[2-(benzoylamino)ethyl]-2-(methoxymethoxy)- (9CI) (CA INDEX NAME)

PAGE 1-B

IT 250207-72-8P

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and complexation with iron and protonation consts.)

250207-72-8 HCAPLUS RN

Benzamide, 4,4',4''-[nitrilotris[2,1-ethanediyl(methylimino)methylene]]tri CN s[N-[2-(benzoylamino)ethyl]-2-hydroxy- (9CI) (CA INDEX NAME)

PAGE 1-A - NH- CH2 CH₂ N-Me CH₂ NH-CH2-CH2-NH-C Me CH₂ CH2-N-CH2-CH2-N-CH2-CH2-

RL: FMU (Formation, unclassified); PRP (Properties); FORM (Formation, nonpreparative) (stability consts.

L33 ANSWER 7 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1999:624043 HCAPLUS

DN 131:358193

ED Entered STN: 01 Oct 1999

TI Structure and switching in bent-shaped molecular liquid crystal systems with two mesogenic groups linked by alkylene spacer

AU Choi, Suk-Won; Zennyoji, Masahito; Takanishi, Yoichi; Takezoe, Hideo; Niori, Teruki; Watanabe, Junji

CS Department of Organic and Polymeric Materials, Tokyo Institute of Technology, Tokyo, 152-8552, Japan

Molecular Crystals and Liquid Crystals Science and Technology, Section A:
Molecular Crystals and Liquid Crystals (1999), 328, 185-192
CODEN: MCLCE9; ISSN: 1058-725X

PB Gordon & Breach Science Publishers

DT Journal

LA English

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 73, 75, 76

AB Structure and switching behavior were studied in liquid crystals consisting of bent-shaped mols. with the linkage of an alkylene spacer, m(O)AMnAM(O)m, where m and n are carbon nos. of end and spacer chains, resp. All the compds. exhibit a fan-shaped texture without a fringe structure in their smectic phase, different from the texture of conventional bent (banana)-shaped liquid crystals. In 12AM5AM12, two switching current peaks indicating the antiferroelec. phase were observed The spontaneous polarization was about 600 nC/cm2. The color change due to the birefringence change is associated with the switching, though the extinction direction between crossed polarizers remains the same. This clearly proves that the bent-mols. do not tilt with respect to the layer normal. In the dielec. measurements, there exist two relaxations at 600 kHz and below 100 Hz and they are suppressed by a biased voltage. Structure and switching in the other systems, 80AM5AM08, are also

described briefly. ST ferroelec antiferroelec switching bent shaped mol liq crystal system IT Liquid crystals (antiferroelec.; structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer in relation to) IT Liquid crystals (ferroelec.; structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer) IT Ferroelectric materials (liquid-crystal; structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer) ΤТ Antiferroelectric materials (liquid-crystal; structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer in relation to) IT Antiferroelectricity Birefringence Refractive index Spontaneous dielectric polarization (structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer) ITDielectric relaxation Ferroelectric switching Liquid crystal displays Periodic structures (structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer in relation to) ITElectrooptical effect (switching; structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer in relation to) IT**250689-35-1**, 80AM5AM08 RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (80AM5AM08; structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer) IT **243975-13-5**, 12AM5AM12 RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer) RE, CNT THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD

- (1) Link, D; Science 1007, V278, P1924
- (2) Nakata, Y; Polymer J 1997, V29, P193 HCAPLUS
- (3) Niori, T; J Mater Chem 1996, V6, P1231 HCAPLUS
- (4) Sekine, T; J Mater Chem 1997, V7, P1307 HCAPLUS
- (5) Sekine, T; Jpn J Appl Phys 1997, V36, P6455 HCAPLUS (6) Sekine, T; Jpn J Appl Phys 1997, V36, PL1201
- (7) Watanabe, J; EKISHO (in Japanese) 1998, V2, P28
- (8) Watanabe, J; Jpn J Appl Phys 1998, V37, PL401 HCAPLUS

IT **250689-35-1**, 80AM5AM08

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(80AM5AM08; structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer)

RN 250689-35-1 HCAPLUS

CN Benzoic acid, 4-[(E)-[[4-(octyloxy)phenyl]imino]methyl]-, 1,5-pentanediyl ester (9CI) (CA INDEX NAME)

Double bond geometry as shown.

PAGE 1-B

IT **243975-13-5**, 12AM5AM12

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(structure and ferroelec./antiferroelec. switching in bent-shaped mol. liquid crystal systems with two mesogenic groups linked by alkylene spacer)

RN 243975-13-5 HCAPLUS

CN Benzoic acid, 4-[(E)-[(4-dodecylphenyl)imino]methyl]-, 1,5-pentanediyl ester (9CI) (CA INDEX NAME)

Double bond geometry as shown.

PAGE 1-A

PAGE 1-B

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L33 ANSWER 8 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN
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AN 1999:597073 HCAPLUS

DN 131:235833

ED Entered STN: 22 Sep 1999

TI Ferroelectric liquid crystal compound for liquid crystal display device

IN Watanabe, Junji; Nakata, Yasukazu

PA Lintec Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C09K019-06

ICS C09K019-22; G02F001-13

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

Section cross-reference(s): 75

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 11256163	A2	19990921	JP 1998-65572	19980316
PRAI	JP 1998-65572		19980316		

AB The ferroelec. liquid crystal compound has structure A-M1-X1-B-X2-M2-A (A = terminal group; M1-2 = mesogen; X1-2 = connecting group; B = flexible group), wherein the flexible group is non-chiral divalent aliphatic group. The ferroelec. liquid crystal compound shows the excellent self-polarization without an optically active group.

ST ferroelec liq crystal display

IT Liquid crystal displays

(ferroelec. liquid crystal compound for liquid crystal display device)

IT Liquid crystals

(ferroelec.; ferroelec. liquid crystal compound for liquid crystal display device)

IT Ferroelectric materials

(liquid-crystal; ferroelec. liquid crystal compound for liquid crystal display

device)

IT 243975-13-5P 243975-14-6P 243975-15-7P 243975-16-8P RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(Ferroelec. liquid crystal compound for liquid crystal display device)

IT 104-42-7, 4-Dodecylaniline 111-29-5, 1,5-Pentanediol 111-46-6, reactions 538-75-0, N,N'-Dicyclohexylcarbodiimide 623-27-8, Terephthalic aldehyde 16245-79-7, 4-Octylaniline 25265-71-8, Dipropylene glycol

RL: RCT (Reactant); RACT (Reactant or reagent)

(Ferroelec. liquid crystal compound for liquid crystal display device)

IT 243975-13-5P

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(Ferroelec. liquid crystal compound for liquid crystal display device)

RN 243975-13-5 HCAPLUS

CN Benzoic acid, 4-[(E)-[(4-dodecylphenyl)imino]methyl]-, 1,5-pentanediyl ester (9CI) (CA INDEX NAME)

Double bond geometry as shown.

PAGE 1-B

L33 ANSWER 9 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

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1999:309715 HCAPLUS
AN
     131:102716
DN
ED
     Entered STN: 21 May 1999
     Analysis of the Structure of Dendrimers in Solution by Small-Angle Neutron
TI
     Scattering Including Contrast Variation
     Poetschke, D.; Ballauff, M.; Lindner, P.; Fischer, M.; Voegtle, F.
ΑU
     Polymer-Institut, Universitaet Karlsruhe, Karlsruhe, 76128, Germany
CS
     Macromolecules (1999), 32(12), 4079-4087
SO
     CODEN: MAMOBX; ISSN: 0024-9297
PB
     American Chemical Society
DT
     Journal
     English
LΑ
     36-2 (Physical Properties of Synthetic High Polymers)
CC
     The anal. of a dendrimer of fifth generation by small-angle neutron
AΒ
     scattering (SANS) in solution is presented. The contrast of the solute
     toward the solvent dimethylacetamide (DMA) is changed systematically by
     measurements of the dendrimer in mixts. of deuterated with protonated DMA.
     Addnl. SANS measurements at highest contrast and varying dendrimer concns.
     allowed to determine the structure factor of the dendrimers in solution SANS
     intensities measured at different contrast are shown to yield the contrast
     .hivin.\rho - \rho m where .hivin.\rho is the average scattering length d.
     of the dissolved dendrimer and pm is the scattering length d. of the
     solvent. This allows to determine the mol. weight of the dendrimer in an
     unambiguous fashion. The comparison of the measured and the calculated mol.
     weight demonstrates that the dendritic structure under consideration here is
     not fully perfect. The anal. of the radial structure of the dendrimer
     rests on the decomposition of the measured intensities into terms depending on
     powers of the contrast .hivin.\rho - \rhom. The leading term which
     scales with the square of the contrast leads to the determination of the
     scattering intensity referring to infinite contrast. This allows to
     elucidate the radial scattering length d. in an unambiguous manner. The
     anal. demonstrates that the present dendrimer, composed of flexible units,
     has a compact structure where the d. has its maximum at the center of the
     mol. This is in accord with recent theor. deductions.
     dendrimer structure neutron scattering contrast variation
ST
     Neutron scattering
IT
     Radius of gyration
     Structure factor
        (anal. of structure of dendrimers in solution by small-angle neutron
        scattering including contrast variation)
IT
     Polyamines
     Polyamines
     RL: PRP (Properties)
        (dendrimers; anal. of structure of dendrimers in solution by small-angle
        neutron scattering including contrast variation)
IT
     Dendritic polymers
     Dendritic polymers
     RL: PRP (Properties)
         (polyamines; anal. of structure of dendrimers in solution by small-angle
        neutron scattering including contrast variation)
                  29530-53-8 115584-73-1D, reaction products with
IT
     poly(propylene imine) dendrimers
                                       179125-43-0D, reaction products with 3
     and 4-(phenylazo)benzoic acid hydroxysuccinimide ester derivs.
     179125-45-2D, reaction products with 3 and 4-(phenylazo)benzoic acid
     hydroxysuccinimide ester derivs.
                                        208049-78-9D, reaction products with
     poly(propylene imine) dendrimers 208049-79-0 208049-80-3
     208049-81-4 208049-82-5
     RL: PRP (Properties)
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(anal. of structure of dendrimers in solution by small-angle neutron scattering including contrast variation)

RE.CNT 29 THERE ARE 29 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Archut, A; Chem Eur J 1998, V4, P699 HCAPLUS
- (2) Banchio, A; Prog Colloid Polym Sci 1998, V110, P54 HCAPLUS
- (3) Bauer, B; Polym Mater Sci Eng 1992, V66, P704
- (4) Boris, D; Macromolecules 1996, V29, P7251 HCAPLUS
- (5) Briber, R; Polym Mater Sci Eng 1992, V67, P430 HCAPLUS
- (6) Cavallo, L; Chem Eur J 1998, V4, P927 HCAPLUS
- (7) de Gennes, P; J Phys (Paris) 1983, V44, PL351
- (8) Dingenouts, N; Adv Polym Sci 1999, V144, P1 HCAPLUS
- (9) Feigin, L; Structure Analysis by Small-Angle X-Ray Scattering and Neutron Scattering 1987
- (10) Gosh, R; A Computing Guide for Small-Angle Scattering Experiments 1998
- (11) Hickl, P; Macromolecules 1996, V29, P4006 HCAPLUS
- (12) Hickl, P; Physica A 1997, V235, P238 HCAPLUS
- (13) Higgins, J; Polymers and Neutron Scattering 1994
- (14) Hummelen, J; Chem Eur J 1997, V3, P1489 HCAPLUS
- (15) Hyman, A; Macromolecules 1975, V8, P849 HCAPLUS
- (16) Issberner, J; Angew Chem Int Ed 1994, V33, P2413
- (17) Lescanec, R; Macromolecules 1990, V23, P2280 HCAPLUS
- (18) Luzatti, V; J Mol Biol 1976, V101, P115
- (19) Mansfield, M; Macromolecules 1993, V26, P4262 HCAPLUS
- (20) Murat, M; Macromolecules 1996, V29, P1278 HCAPLUS
- (21) Newkome, G; Dendritic Molecules 1996
- (22) Prosa, T; J Polym Sci, Part B: Polym Phys 1997, V35, P2913 HCAPLUS
- (23) Ragnetti, M; Colloid Polym Sci 1986, V264, P32 HCAPLUS
- (24) Ramzi, A; Macromolecules 1998, V31, P1621 HCAPLUS
- (25) Scherrenberg, R; Macromolecules 1998, V31, P456 HCAPLUS
- (26) Stuhrmann, H; Z Phys Chem (Munich) 1967, V56, P334 HCAPLUS
- (27) Tomalia, D; Angew Chem Int Ed 1990, V29, P138
- (28) Uppuluri, S; Macromolecules 1998, V31, P4498 HCAPLUS
- (29) Welch, P; Macromolecules 1998, V31, P5892 HCAPLUS
- TT 208049-79-0 208049-80-3 208049-81-4 208049-82-5
 - RL: PRP (Properties)
 - (anal. of structure of dendrimers in solution by small-angle neutron scattering including contrast variation)
- RN 208049-79-0 HCAPLUS
- CN Benzamide, N,N',N'',N'''-[1,2-ethanediylbis(nitrilodi-3,1-propanediyl)]tetrakis[4-(phenylazo)- (9CI) (CA INDEX NAME)

$$\begin{array}{c} Ph-N=N \\ \hline \\ C=0 \\ NH \\ (CH_2)_3 \\ N-(CH_2)_3-NH-C \\ \hline \\ CH_2 \\ CH_2 \\ CH_2 \\ CH_2 \\ CH_2 \\ O \\ N=N-Ph \\ N=N-Ph \\ \end{array}$$

RN 208049-80-3 HCAPLUS
CN Benzamide, N,N',N'',N'''-[1,2-ethanediylbis(nitrilodi-3,1-propanediyl)]tetrakis[3-(phenylazo)- (9CI) (CA INDEX NAME)

RN 208049-81-4 HCAPLUS
CN Benzamide, N,N',N'',N''',N'''',N''''',N''''',N''''',N'''''-[1,2-ethanediylbis[nitrilobis(3,1-propanediylnitrilodi-3,1-propanediyl)]]octakis[4-(phenylazo)- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

RN 208049-82-5 HCAPLUS
CN Benzamide, N,N',N'',N''',N'''',N''''',N''''',N'''''-[1,2-ethanediylbis[nitrilobis(3,1-propanediylnitrilodi-3,1-propanediyl)]]octakis[3-(phenylazo)-(9CI) (CA INDEX NAME)

PAGE 1-A

$$Ph-N=N$$
 $C=0$
 NH
 $CH_2)_3$
 $N-(CH_2)_3-NH=0$
 $CH_2)_3$
 $CH_2)_3$

PAGE 1-B

PAGE 2-B

$$(CH_2)_3$$
 $-(CH_2)_3 - N - (CH_2)_3 - NH - C$
 $N = N - Ph$
 $CH_2)_3 - NH - C$
 $N = N - Ph$
 $N = N - Ph$

L33 ANSWER 10 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1998:759204 HCAPLUS

DN 130:82105

ED Entered STN: 04 Dec 1998

TI Toward Photoswitchable Dendritic Hosts. Interaction between Azobenzene-Functionalized Dendrimers and Eosin

AU Archut, Andreas; Azzellini, Gianluca Camillo; Balzani, Vincenzo; Cola, Luisa; Voegtle, Fritz

CS Kekule-Institut fuer Organische Chemie und Biochemie, Universitaet Bonn, Bonn, D-53121, Germany

SO Journal of the American Chemical Society (1998), 120(47), 12187-12191 CODEN: JACSAT; ISSN: 0002-7863

PB American Chemical Society

DT Journal

LA English

CC 36-5 (Physical Properties of Synthetic High Polymers)

AB Two poly(propyleneimine) dendrimers bearing up to 32 photoisomerizable azobenzene groups in the periphery have been used as potential hosts for eosin Y (2',4',5',7'-tetrabromofluorescein dianion). The all-E azobenzene

dendrimers can be reversibly switched to their Z form by light excitation. Both the E and Z forms of the dendrimers quench the eosin fluorescence by a static mechanism. The quenching is most likely due to an electron-transfer reaction between the singlet excited state of eosin and the tertiary amine units present along the branches of the dendrimers. Quenching by the Z form of the dendrimers is more efficient than quenching by the E form. The E \rightarrow Z and Z \rightarrow E photoisomerization reactions of the azobenzene units of the dendrimers are sensitized by eosin via a triplet-triplet energy transfer mechanism. The results obtained indicate that eosin is hosted by the dendrimers and suggest that the Z forms are more efficient hosts than the E forms.

ST interaction azobenzene functionalized dendrimer eosin; photoisomerization azobenzene functionalized dendrimer eosin interaction

IT Electron transfer

Energy transfer

Excited state

Fluorescence

(interaction between photoswitchable azobenzene-functionalized dendrimers and eosin)

IT Dendritic polymers

RL: PRP (Properties)

(interaction between photoswitchable azobenzene-functionalized dendrimers and eosin)

IT Isomerization

(photoisomerization; interaction between photoswitchable azobenzene-functionalized dendrimers and eosin)

IT 107-13-1D, Acrylonitrile, hydrogenated, Michael addition dendrimers 17372-87-1, Eosin Y 29530-52-7D, Benzamide, m-(phenylazo)-, Michael addition dendrimers 29530-53-8D, Benzamide, p-(phenylazo)-, Michael addition dendrimers 208049-79-0D, Michael addition dendrimers RL: PRP (Properties)

(interaction between photoswitchable azobenzene-functionalized dendrimers and eosin)

IT 107-13-1D, 2-Propenenitrile, hydrogenated, Michael-addition dendrimers, properties

RL: PRP (Properties)

(polypropylenimine; interaction between photoswitchable azobenzene-functionalized dendrimers and eosin)

RE.CNT 32 THERE ARE 32 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Archut, A; Chem Eur J 1998, V4, P669
- (2) Ashton, P; Chem Eur J 1997, V3, P974 HCAPLUS
- (3) Balzani, V; Supramolecular Photochemistry 1991
- (4) Bartholomeu, R; J Chem Soc (C) 1971, P2347
- (5) Bortolus, P; J Phys Chem 1979, V83, P648 HCAPLUS
- (6) Buhleier, E; Synthesis 1978, P155 HCAPLUS
- (7) Credi, A; Spectrochim Acta A 1998, V54, P159
- (8) Frechet, J; Science 1994, V263, P1710 HCAPLUS
- (9) Gilbert, A; Essentials of Molecular Photochemistry 1991
- (10) Gorman, C; Adv Mater 1998, V10, P295 HCAPLUS
- (11) Hummelen, J; Chem Eur J 1997, V3, P1493
- (12) Jansen, J; J Am Chem Soc 1995, V117, P4417 HCAPLUS
- (13) Jansen, J; Science 1994, V266, P1226 HCAPLUS
- (14) Jiang, D; Nature 1997, V388, P454 HCAPLUS
- (15) Junge, D; Chem Commun 1997, P857 HCAPLUS
- (16) Kumar, G; Chem Rev 1989, V89, P1915 HCAPLUS
- (17) Lakowicz, J; Principles of Fluorescence Spectroscopy 1986
- (18) Mann, C; Electrochemical Reactions in Non-aqueous Systems 1970

(19) Miklis, P; J Am Chem Soc 1997, V119, P7458 HCAPLUS

(20) Moser, J; J Am Chem Soc 1984, V106, P6567

(21) Murov, S; Handbook of Photochemistry 1993

(22) Naylor, A; J Am Chem Soc 1989, V111, P2339 HCAPLUS

(23) Newkome, G; Dendritic Macromolecules: Concepts, Syntheses, Perspectives

(24) Rau, H; Photochromism Molecules and Systems 1990

(25) Ronayette, J; Can J Chem 1974, V52, P1858 HCAPLUS

(26) Scandola, F; Comprehensive Supramolecular Chemistry 1996, V10, P687

(27) Shinkai, S; Comprehensive Supramolecular Chemistry 1996, V1, P671 HCAPLUS

(28) Shinkai, S; Top Curr Chem 1984, V121, P67 HCAPLUS

(29) Tomalia, D; Top Curr Chem 1993, V165, P193 HCAPLUS

(30) Turro, C; J Phys Chem 1995, V99, P5512

(31) Turro, N; Modern Molecular Photochemistry 1978

(32) Valdes-Aguilera, O; Acc Chem Res 1989, V22, P171 HCAPLUS

IT 208049-79-0D, Michael addition dendrimers

RL: PRP (Properties)

(interaction between photoswitchable azobenzene-functionalized dendrimers and eosin)

RN 208049-79-0 HCAPLUS

CN Benzamide, N,N',N'',N'''-[1,2-ethanediylbis(nitrilodi-3,1-propanediyl)]tetrakis[4-(phenylazo)- (9CI) (CA INDEX NAME)

$$\begin{array}{c} Ph-N=N \\ \hline \\ C=0 \\ \hline \\ NH \\ \hline \\ (CH_2)_3 \\ \hline \\ N-(CH_2)_3-NH-C \\ \hline \\ CH_2 \\ \hline \\ O \\ \hline \\ N=N-Ph \\ \hline$$

L33 ANSWER 11 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1998:308903 HCAPLUS

DN 129:34339

ED Entered STN: 28 May 1998

TI Azobenzene-functionalized cascade molecules: photoswitchable supramolecular systems

AU Archut, Andreas; Vogtle, Fritz; De Cola, Luisa; Azzellini, Gianluca Camillo; Balzani, Vincenzo; Ramanujam, P. S.; Berg, Rolf H.

CS Kekule-Inst. Organische Chem. & Biochemie, Univ. Bonn, Bonn, D-53121,
Germany

SO Chemistry--A European Journal (1998), 4(4), 699-706

CODEN: CEUJED; ISSN: 0947-6539 Wiley-VCH Verlag GmbH PB Journal DTLΑ English CC 74-1 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes) Section cross-reference(s): 38, 73 Cascade mols. bearing up to 32 azobenzene groups in the periphery have AΒ been prepared from poly(propylene imine) dendrimers and N-hydroxysuccinimide esters. The dendritic azobenzene species show similar isomerization properties as the corresponding azobenzene monomers. The all-E azobenzene dendrimer units can be reversibly switched to the Z form by light of the appropriate wavelength and can be converted back to the E form by either irradiation or by heating. That the photoisomerization quantum yield of each photoactive unit is not dependent on the number of such units present in the species shows there is so far no effective steric constraint towards photoisomerism on increasing dimension (generation) of the dendrimer. The first attempts to use dendrimers for holog. materials are described: It is shown that holog. gratings with diffraction efficiencies up to about 20% can be optically recorded in thin films of azobenzene dendrimers. STazo compd dendrimer holog data storage; photochem supramol chem optical recording ΙT Holographic diffraction gratings Holography Optical recording materials Optical switches Supramolecular structure (azobenzene-functionalized cascade mols. for photoswitchable supramol. systems) ITIsomerization (cis-trans, photochem.; azobenzene-functionalized cascade mols. for photoswitchable supramol. systems) IT Polyamines Polvamines RL: PNU (Preparation, unclassified); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (dendrimers, from divergent approach; azobenzene-functionalized cascade mols. for photoswitchable supramol. systems) IT Dendritic polymers Dendritic polymers RL: PNU (Preparation, unclassified); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyamines, from divergent approach; azobenzene-functionalized cascade mols. for photoswitchable supramol. systems) ΙT 29530-52-7 29530-53-8 RL: PRP (Properties) (azobenzene-functionalized cascade mols. for photoswitchable supramol. systems) 115584-73-1DP, reaction products with poly(propylene imine) dendrimers IT 179125-43-0DP, reaction products with 3 and 4-(phenylazo)benzoic acid hydroxysuccinimide ester derivs. 179125-45-2DP, reaction products with 3 and 4-(phenylazo)benzoic acid hydroxysuccinimide ester derivs. 208049-78-9DP, reaction products with poly(propylene imine) dendrimers RL: PNU (Preparation, unclassified); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (dendritic and oligomeric; from divergent approach; azobenzenefunctionalized cascade mols. for photoswitchable supramol. systems)

208049-80-3P 208049-81-4P

TT

RL: PNU (Preparation, unclassified); PRP (Properties); PREP (Preparation) (from divergent approach; azobenzene-functionalized cascade mols. for photoswitchable supramol. systems)

IT 208049-79-0P 208049-82-5P

RL: PNU (Preparation, unclassified); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(from divergent approach; azobenzene-functionalized cascade mols. for photoswitchable supramol. systems)

RE.CNT 59 THERE ARE 59 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Anspon, H; Org Synth 1945, V25, P86 HCAPLUS
- (2) Ashton, P; Chem Eur J 1997, V3, P974 HCAPLUS
- (3) Balzani, V; Supramolecular Photochemistry 1991
- (4) Balzani, V; Supramolecular Photochemistry, chapter 7 1991
- (5) Bauer, M; Chem Ber 1992, V125, P1675 HCAPLUS
- (6) Bencini, A; J Chem Soc Perkin Trans 2 1998, P413 HCAPLUS
- (7) Berg, R; Nature 1996, V383, P505 HCAPLUS
- (8) Brauchle, C; Adv Mater 1991, V3, P420
- (9) Buhleier, E; Synthesis 1978, P155 HCAPLUS
- (10) Coleman, G; Org Synth 1945, V25, P87 HCAPLUS
- (11) de Brabander-van Den Berg, E; Angew Chem 1993, V105, P1370 HCAPLUS
- (12) de Brabander-van Den Berg, E; Angew Chem, Int Ed Engl 1993, V32, P1308
- (13) Drexler, K; Nanosystems 1992
- (14) Fischer, E; EPA Newsletter 1984, V21, P33
- (15) Freemantle, M; Chem Eng News 1997, VMay 26, P30
- (16) Freundler, P; Bull Soc Chim Fr 1907, V1(4), P221
- (17) Hawker, C; J Am Chem Soc 1990, V112, P7638 HCAPLUS
- (18) Holme, N; Appl Phys Lett 1995, V70, P1518
- (19) Hummelen, J; Chem Eur J 1997, V3, P1489 HCAPLUS
- (20) Ikeda, T; Science 1995, V268, P1873 HCAPLUS
- (21) Issberner, J; Angew Chem 1994, V106, P2507 HCAPLUS
- (22) Issberner, J; Angew Chem Int Ed Engl 1994, V33, P2413
- (23) Jacobson, P; Liebigs Ann 1909, V367, P329
- (24) Jiang, D; Nature 1997, V388, P454 HCAPLUS
- (25) Junge, D; Chem Commun 1997, P857 HCAPLUS
- (26) Kim, C; Bull Korean Chem Soc 1997, V18, P164 HCAPLUS
- (27) Kim, D; Appl Phys Lett 1995, V66, P1166 HCAPLUS
- (28) Kumar, G; Chem Rev 1989, V89, P1915 HCAPLUS
- (29) Lange, P; Inorg Chem 1996, V35, P637 HCAPLUS
- (30) Lartigue, M; Chem Eur J 1996, V2, P1417 HCAPLUS
- (31) Launay, N; Angew Chem 1994, V106, P1682 HCAPLUS
- (32) Launay, N; Angew Chem Int Ed Engl 1994, V33, P1589
- (33) Lehn, J; Supramolecular Chemistry 1995
- (34) Lorenz, K; Macromolecules 1995, V28, P6657 HCAPLUS
- (35) Maier, W; Z Naturforsch A 1959, V14, P882
- (36) Mekelburger, H; Chem Ber 1993, V126, P1161 HCAPLUS
- (37) Moors, R; Chem Ber 1993, V126, P2133 HCAPLUS
- (38) Newkome, G; Dendritic Molecules: Concepts, Syntheses, Perspectives 1996
- (39) Pederson, T; Phys Rev Lett 1998, V80, P89
- (40) Ramanujam, P; Appl Phys Lett 1996, V68, P1329 HCAPLUS
- (41) Ranjit, K; Angew Chem 1997, V109, P91
- (42) Ranjit, K; Angew Chem Int Ed Engl 1997, V36, P147 HCAPLUS
- (43) Rau, H; Photochromism, Molecules and Systems, chapter 4 1990
- (44) Rochon, P; Appl Phys Lett 1995, V66, P136 HCAPLUS
- (45) Serroni, S; J Mater Chem 1997, V7, P1227 HCAPLUS
- (46) Seyferth, D; Organometallics 1994, V13, P2682 HCAPLUS
- (47) Shinkai, S; Comprehensive Supramolecular Chemistry 1996
- (48) Shinkai, S; J Chem Soc Perkin Trans 2 1982, P1261 HCAPLUS

- (49) Shinkai, S; Topics Curr Chem 1984, V121, P67 HCAPLUS
- (50) Tomalia, D; Macromolecules 1986, V19, P2466 HCAPLUS
- (51) Tomalia, D; Polym J 1985, V17, P117 HCAPLUS
- (52) van der Made, A; Adv Mater 1993, V5, P466 HCAPLUS
- (53) van der Made, A; J Chem Soc Chem Commun 1992, P1400 HCAPLUS
- (54) Vennen, H; Hoechst High Chem Magazin 1994, V15, P57
- (55) Wooley, K; J Am Chem Soc 1991, V113, P4252 HCAPLUS
- (56) Worner, C; Angew Chem 1993, V105, P1367
- (57) Worner, C; Angew Chem Int Ed Engl 1993, V32, P1306
- (58) Wurthner, F; Angew Chem 1995, V107, P503
- (59) Wurthner, F; Angew Chem Int Ed Engl 1995, V34, P446
- IT 208049-80-3P 208049-81-4P

RL: PNU (Preparation, unclassified); PRP (Properties); PREP (Preparation) (from divergent approach; azobenzene-functionalized cascade mols. for photoswitchable supramol. systems)

- RN 208049-80-3 HCAPLUS
- CN Benzamide, N,N',N'',N'''-[1,2-ethanediylbis(nitrilodi-3,1-propanediyl)]tetrakis[3-(phenylazo)- (9CI) (CA INDEX NAME)

$$\begin{array}{c}
C = 0 \\
NH \\
(CH_2)_3 & 0 \\
N = N - Ph \\
N - (CH_2)_3 - NH - C
\end{array}$$

$$\begin{array}{c}
CH_2 \\
CH_2 \\
CH_2
\end{array}$$

$$\begin{array}{c}
CH_2 \\
CH_2$$

$$CH_2 \\
CH_2$$

$$CH_2$$

$$C$$

- RN 208049-81-4 HCAPLUS
- CN Benzamide, N,N',N'',N''',N'''',N''''',N''''',N'''''-[1,2-ethanediylbis[nitrilobis(3,1-propanediylnitrilodi-3,1-propanediyl)]]octakis[4-(phenylazo)- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

PAGE 2-B

IT 208049-79-0P 208049-82-5P

RL: PNU (Preparation, unclassified); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(from divergent approach; azobenzene-functionalized cascade mols. for photoswitchable supramol. systems)

RN 208049-79-0 HCAPLUS

CN Benzamide, N,N',N'',N'''-[1,2-ethanediylbis(nitrilodi-3,1-propanediyl)]tetrakis[4-(phenylazo)- (9CI) (CA INDEX NAME)

$$\begin{array}{c}
C = 0 \\
NH \\
(CH_2)_3 & 0 \\
N = N - Ph \\
N - (CH_2)_3 - NH - C
\end{array}$$

$$\begin{array}{c}
N = N - Ph \\
CH_2 & 0 \\
CH_2 & 0
\end{array}$$

$$\begin{array}{c}
CH_2 & 0 \\
CH_2 & 0
\end{array}$$

$$\begin{array}{c}
CH_2 & 0 \\
CH_2 & 0
\end{array}$$

$$\begin{array}{c}
N = N - Ph \\
CH_2 & 0
\end{array}$$

$$\begin{array}{c}
N = N - Ph
\end{array}$$

RN 208049-82-5 HCAPLUS
CN Benzamide, N,N',N'',N''',N'''',N''''',N'''''-[1,2-ethanediylbis[nitrilobis(3,1-propanediylnitrilodi-3,1-propanediyl)]]octakis[3-(phenylazo)-(9CI) (CA INDEX NAME)

PAGE 2-B

$$(CH_2)_3$$
 $-(CH_2)_3 - N - (CH_2)_3 - N + C$
 0
 $N = N - Ph$
 0
 $N = N - Ph$
 0

L33 ANSWER 12 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1997:550609 HCAPLUS

DN 127:262416

ED Entered STN: 29 Aug 1997

TI Molecular meccano. 10. Toward controllable molecular shuttles

AU Anelli, Pier-Lucio; Asakawa, Masumi; Ashton, Peter R.; Bissell, Richard A.; Clavier, Gilles; Gorski, Romuald; Kaifer, Angel E.; Langford, Steven J.; Mattersteig, Gunter; Menzer, Stephan; Philp, Douglas; Slawin, Alexandra M. Z.; Spencer, Neil; Stoddart, J. Fraser; Tolley, Malcolm S.; Williams, David J.

CS Sch. Chem., Univ. Birmingham, Edgbaston, Birmingham, B15 2TT, UK

SO Chemistry--A European Journal (1997), 3(7), 1113-1135 CODEN: CEUJED; ISSN: 0947-6539

PB Wiley-VCH

DT Journal

LA English

CC 22-13 (Physical Organic Chemistry) Section cross-reference(s): 75

AΒ A number of nanometer-scale mol. assemblies, based on rotaxane-type structures, have been synthesized by means of a template-directed strategy from simple building blocks that, on account of the mol. recognition arising from the noncovalent interactions between them, are able to self-assemble into potential mol. abacuses. In all the cases investigated, the π -electron-deficient tetracationic cyclophane cyclobis(para-quat-p-phenylene) is constrained mech. around a dumbbell-shaped component consisting of a linear polyether chain intercepted by at least two, if not three, π -electron-rich units and terminated at each end by blocking groups or stoppers. The development of an approach toward constructing these mol. abacuses, in which the tetracationic cyclophane is able to shuttle back and forth with respect to the dumbbell-shaped component, begins with the self-assembly of a [2]rotaxane consisting of two hydroquinone rings sym. positioned within a polyether chain terminated by triisopropylsilyl ether blocking groups. In this first so called mol. shuttle, the tetracationic cyclopane oscillates in a degenerate fashion between the two π -electron-rich hydroquinone rings. Replacement of one of the hydroquinone rings-or the insertion of another π -electron-rich ring system between the two hydroquinine rings-introduces the possibility of translational isomerism, a phenomenon that arises because of the different relative positions and populations of the tetracationic cyclophane with respect to the π -donor sites on the dumbbell-shaped component. In two subsequent [2] rotaxanes, one of the hydroquinone rings in the dumbbell-shaped component is replaced, first by a p-xylyl and then by an indole unit. Finally, a tetrathiafulvalene (TTF)

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unit is positioned between two hydroquinone rings in the dumbbell-shaped
     component. Spectroscopic and electrochem, investigations carried out on
     these first-generation mol. shuttles show that they could be developed as
     mol. switches.
     kinetics translational isomerism mol shuttle rotaxanes; mol shuttle
     rotaxanes electrochem UV NMR
     Crystal structure
     Molecular structure
        (crystallog. of complex between cyclobis(paraquat-p-phenylene) and
        2-methylindole)
     Nomenclature, general
        (mol. shuttles)
     Activation energy
     Binding energy
     Cyclic voltammetry
     Electrostatic potential
     Hydrogen bond
     NMR (nuclear magnetic resonance)
     Oxidation, electrochemical
     Physical process kinetics
     Reduction, electrochemical
     UV and visible spectra
        (prepn.and translational isomerism of controllable mol. shuttles)
     Rotaxanes
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN
     (Synthetic preparation); PREP (Preparation); PROC (Process)
        (prepn.and translational isomerism of controllable mol. shuttles)
     Isomers
        (translation; prepn.and translational isomerism of controllable mol.
        shuttles)
     145839-50-5P
     RL: BYP (Byproduct); PREP (Preparation)
        (byproduct; preparation of controllable mol. shuttles)
     147553-58-0P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (crystallog. of complex between cyclobis(paraguat-p-phenylene) and
        2-methylindole)
     5978-08-5P 134261-30-6P
                                 134286-52-5P 145839-35-6P
     145839-36-7P
                    145839-37-8P
                                   145839-38-9P
                                                 145839-39-0P
                                                                  145839-40-3P
     145839-41-4P
                    145839-42-5P
                                   145839-43-6P
                                                  145839-44-7P
                                                                  145839-45-8P
                    145839-47-0P
     145839-46-9P
                                   145839-48-1P
                                                  145839-49-2P
                                                                  145995-58-0P
     146063-36-7P
                    146063-42-5P
                                   146063-43-6P
                                                  146063-44-7P
                                                                  146063-45-8P
     146063-46-9P
                    146063-47-0P
                                   146063-48-1P
                                                  195878-86-5P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (intermediate; preparation of controllable mol. shuttles)
     134286-51-4P
                    145995-61-5P
                                  195878-85-4P 195878-87-6P
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN
     (Synthetic preparation); PREP (Preparation); PROC (Process)
        (prepn.and translational isomerism of controllable mol. shuttles)
     623-24-5, 1,4-Bis(bromomethyl)benzene 5292-43-3, tert-Butyl bromoacetate
     51145-58-5, 4-Benzyloxyphenylhydrazine
                                              108861-20-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reagent; preparation of controllable mol. shuttles)
     4141-19-9
                 4141-20-2
                            5197-62-6
                                        14556-10-6
                                                      134881-72-4
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (starting material; preparation of controllable mol. shuttles)
RE.CNT
       127
              THERE ARE 127 CITED REFERENCES AVAILABLE FOR THIS RECORD
```

RE

- (1) Abboud, J; Progr Phys Org Chem 1981, V13, P485
- (2) Amabilino, D; Chem Rev 1995, V95, P2725 HCAPLUS
- (3) Amabilino, D; J Am Chem Soc 1995, V117, P1271 HCAPLUS
- (4) Amabilino, D; J Am Chem Soc 1996, V118, P3905 HCAPLUS
- (5) Amabilino, D; J Chem Soc Chem Commun 1994, P2475 HCAPLUS
- (6) Amabilino, D; New Scientist 1994, V141(1913), P25 HCAPLUS
- (7) Amabilino, D; Pure Appl Chem 1993, V65, P2351 HCAPLUS
- (8) Anelli, P; J Am Chem Soc 1991, V113, P5131 HCAPLUS
- (9) Anelli, P; J Am Chem Soc 1992, V114, P193 HCAPLUS
- (10) Anon; UltraFit 1992
- (11) Asakawa, M; J Am Chem Soc 1997, V119, P302 HCAPLUS
- (12) Asakawa, M; J Org Chem 1996, V61, P9591 HCAPLUS
- (13) Ashton, P; Angew Chem Int Ed Engl 1988, V27, P1550
- (14) Ashton, P; Angew Chem Int Ed Engl 1989, V28, P1396
- (15) Ashton, P; Angew Chem Int Ed Engl 1995, V34, P571 HCAPLUS
- (16) Ashton, P; Chem Eur J 1996, V2, P729 HCAPLUS
- (17) Ashton, P; J Am Chem Soc 1995, V117, P11171
- (18) Ashton, P; J Am Chem Soc 1996, V118, P4931 HCAPLUS
- (19) Ashton, P; J Chem Soc Chem Commun 1992, P1125
- (20) Ashton, P; J Chem Soc Chem Commun 1993, P1269 HCAPLUS
- (21) Ashton, P; J Chem Soc Chem Commun 1993, P1274 HCAPLUS
- (22) Ashton, P; J Chem Soc Chem Commun 1994, P177 HCAPLUS
- (23) Ashton, P; Synlett 1992, P914 HCAPLUS
- (24) Ashton, P; Synlett 1992, P919 HCAPLUS
- (25) Ashton, P; Synlett 1992, P923 HCAPLUS
- (26) Ashton, P; Tetrahedron Lett 1991, V32, P6235 HCAPLUS
- (27) Ashton, P; Tetrahedron Lett 1996, V37, P6217 HCAPLUS
- (28) Ballardini, R; Angew Chem Int Ed Engl 1993, V32, P1301
- (29) Ballardini, R; Angew Chem Int Ed Engl 1996, V35, P978 HCAPLUS
- (30) Becher, J; J Chem Soc Perkin Trans 1 1990, P175 HCAPLUS
- (31) Benniston, A; J Am Chem Soc 1995, V117, P5275 HCAPLUS
- (32) Bissell, R; Chem Rev Soc 1992, V21, P187 HCAPLUS
- (33) Bissell, R; Computations for the Nano-Scale, NATO ASI Series 1993, V240, P141
- (34) Bissell, R; Nature (London) 1994, V369, P133 HCAPLUS
- (35) Born, M; Angew Chem Int Ed Engl 1995, V34, P309 HCAPLUS
- (36) Brown, C; Synlett 1991, P459 HCAPLUS
- (37) Brown, C; Synlett 1991, P462 HCAPLUS
- (38) Bryce, M; Tetrahedron Lett 1991, V32, P6033 HCAPLUS
- (39) Carter, F; Molecular Electronic Devices 1988
- (40) Carter, F; Science 1991, V254, P1300
- (41) Chambron, J; J Am Chem Soc 1993, V115, P12378 HCAPLUS
- (42) Chambron, J; J Am Chem Soc 1993, V115, P6109 HCAPLUS
- (43) Chambron, J; J Chem Soc Chem Commun 1992, P1131 HCAPLUS
- (44) Chapman, R; J Am Chem Soc 1995, V117, P9081 HCAPLUS
- (45) Clapp, P; J Am Chem Soc 1994, V116, P9166 HCAPLUS
- (46) Colquhoun, H; J Chem Soc Perkin Trans 2 1985, P607 HCAPLUS
- (47) Co'Rdova, E; J Org Chem 1995, V60, P1033 HCAPLUS(48) Credi, A; J Am Chem Soc 1997, V119, P2679 HCAPLUS
- (49) Diederich, F; J Chem Soc Chem Commun 1995, P781 HCAPLUS
- (50) Dietrich-Buchecker, C; Supramolecular Chemistry 1992, P259 HCAPLUS
- (51) Dietrich-Buchecker, C; Tetrahedron Lett 1983, V24, P5095 HCAPLUS
- (52) Drexler, K; Annu Rev Biophys Biomol Struct 1994, V23, P377 HCAPLUS
- (53) Drexler, K; Engines of Creation 1990
- (54) Drexler, K; Nanosystems, Molecular Machinery, Manufacturing and Computation 1992
- (55) Feringa, B; J Am Chem Soc 1991, V113, P5468 HCAPLUS

- (56) Feringa, B; Tetrahedron 1993, V49, P8267 HCAPLUS (57) Freeman, H; Dyes Pigm 1991, V17, P83 HCAPLUS (58) Gibson, H; J Am Chem Soc 1995, V117, P852 HCAPLUS
- (59) Girmay, B; J Chem Soc Chem Commun 1989, P1406 HCAPLUS (60) Girmay, B; J Chem Soc Perkin Trans 1 1992, P383 HCAPLUS
- (61) Goodnow, T; J Am Chem Soc 1991, V113, P4335 HCAPLUS
- (62) Gupta, S; J Org Chem 1959, V24, P409 HCAPLUS
- (63) Hanessian, S; J Am Chem Soc 1995, V117, P7630 HCAPLUS
- (64) Harada, A; J Am Chem Soc 1994, V116, P3192 HCAPLUS
- (65) Harada, A; Nature (London) 1992, V356, P325 HCAPLUS
- (66) Harada, A; Nature (London) 1994, V370, P126 HCAPLUS
- (67) Jager, R; Liebigs Ann Chem 1996, P1201
- (68) James, T; J Am Chem Soc 1995, V117, P8982 HCAPLUS
- (69) Jorgensen, M; J Org Chem 1993, V58, P2785
- (70) Kamlet, M; J Am Chem Soc 1977, V99, P6027 HCAPLUS
- (71) Kamlet, M; J Am Chem Soc 1977, V99, P8325
- (72) Kawai, S; J Chem Soc Chem Commun 1994, P1011 HCAPLUS
- (73) Kelly, T; J Am Chem Soc 1994, V116, P3657 HCAPLUS
- (74) Kern, J; Adv Mater 1996, V8, P580 HCAPLUS
- (75) Kolchinski, A; J Chem Soc Chem Commun 1995, P1289 HCAPLUS
- (76) Krief, A; Tetrahedron 1986, V42, P1209 HCAPLUS
- (77) Lehn, J; Angew Chem Int Ed Engl 1990, V29, P1304
- (78) Li, Z; Angew Chem Int Ed Engl 1995, V34, P2524 HCAPLUS
- (79) Li, Z; Chem Commun 1996, P639 HCAPLUS
- (80) Li, Z; Chem Eur J 1996, V2, P624 HCAPLUS
- (81) Lindsey, J; New J Chem 1991, V15, P153 HCAPLUS
- (82) Livoreil, A; J Am Chem Soc 1994, V116, P9399 HCAPLUS
- (83) Marsella, M; J Am Chem Soc 1994, V116, P9347 HCAPLUS
- (84) Medina, J; J Am Chem Soc 1992, V114, P10583 HCAPLUS
- (85) Mirzoian, A; J Org Chem 1995, V60, P8093 HCAPLUS
- (86) Mizuno, M; J Chem Soc Chem Commun 1978, P18 HCAPLUS
- (87) Muller, W; Science 1993, V262, P1706 MEDLINE
- (88) Myers, R; Anal Chem 1969, V41, P980 HCAPLUS
- (89) Nicholson, R; Anal Chem 1964, V36, P706 HCAPLUS
- (90) Nishio, M; Tetrahedron 1995, V51, P8665 HCAPLUS
- (91) Odell, B; Angew Chem Int Ed Engl 1988, V27, P1547
- (92) Perrin, D; Purification of Laboratory Chemicals, 3rd ed 1988
- (93) Philp, D; Angew Chem Int Ed Engl 1996, V35, P1154
- (94) Philp, D; J Chem Soc Chem Commun 1991, P1584 HCAPLUS
- (95) Philp, D; Synlett 1991, P445 HCAPLUS
- (96) Preece, J; The Ultimate Limits of Fabrication and Measurement 1995, P1
- (97) Pysh, E; J Am Chem Soc 1963, V85, P2124
- (98) Richardson, D; Inorg Chem 1981, V20, P1278 HCAPLUS
- (99) Ringsdorf, H; Angew Chem Int Ed Engl 1988, V27, P113
- (100) Robinson, B; Chem Rev 1963, V63, P373 (101) Robinson, B; Chem Rev 1969, V69, P227 HCAPLUS
- (102) Robinson, B; The Fischer Indole Synthesis 1982
- (103) Sandstrom, J; Dynamic NMR Spectroscopy, Chapter 6 1982
- (104) Sauvage, J; Acc Chem Res 1990, V23, P319 HCAPLUS
- (105) Schill, G; Catenanes, Rotaxanes, and Knots 1971
- (106) Schukat, G; Sulfur Reports 1987, V7, P155 HCAPLUS
- (107) Solladie, N; Angew Chem Int Ed Engl 1996, V35, P906 HCAPLUS
- (108) Steimecke, G; Phosphorus Sulfur 1979, V7, P49 HCAPLUS
- (109) Stoddart, J; Chem Aust 1992, V59, P576 HCAPLUS
- (110) Sundberg, R; The Chemistry of Indoles 1970
- (111) Vance, D; J Am Chem Soc 1994, V116, P9397 HCAPLUS
- (112) Varma, K; Synthesis 1987, P837 HCAPLUS
- (113) Verkade, P; Rec Trav Chim Pays-Bas 1942, V61, P831 HCAPLUS

- (114) Vogtle, F; Chem Eur J 1996, V2, P640
- (115) Vogtle, F; Liebegs Ann Chem 1995, P739
- (116) Vogtle, F; Liebigs Ann Chem 1996, P921
- (117) Vogtle, F; Pure Appl Chem 1996, V68, P225
- (118) Vogtle, F; Synthesis 1996, P353
- (119) Wenz, G; Angew Chem Int Ed Engl 1992, V31, P197
- (120) Wenz, G; Angew Chem Int Ed Engl 1994, P803
- (121) Whitesides, G; Acc Chem Res 1995, V28, P37 HCAPLUS

Page 52

- (122) Whitesides, G; Science 1991, V254, P1312 HCAPLUS
- (123) Wu, C; Chem Mater 1991, V3, P569 HCAPLUS
- (124) Wudl, F; J Org Chem 1974, V39, P3608 HCAPLUS
- (125) Yamaguchi, I; J Am Chem Soc 1996, V118, P1811 HCAPLUS
- (126) Zhao, D; J Org Chem 1991, V56, P3001 HCAPLUS
- (127) Zhou, Q; J Am Chem Soc 1995, V117, P7017 HCAPLUS

IT 145839-35-6P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(intermediate; preparation of controllable mol. shuttles)

RN 145839-35-6 HCAPLUS

CN 1H-Indole, 2-methyl-5-[2-[2-[4-(triphenylmethyl)phenoxy]ethoxy]ethoxy]e thoxy]-3-[2-[2-[2-[4-[2-[2-[4-(triphenylmethyl)phenoxy]ethoxy

PAGE 1-B

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PAGE 1-C

IT 195878-87-6P

RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)

(prepn.and translational isomerism of controllable mol. shuttles)

RN 195878-87-6 HCAPLUS

CN 5,12,19,26-Tetraazoniaheptacyclo[24.2.2.22,5.27,10.212,15.216,19.221,24]te traconta-2,4,7,9,12,14,16,18,21,23,26,28,29,31,33,35,37,39-octadecaene, tetrakis[hexafluorophosphate(1-)], rotaxane compd. with 2-methyl-5-[2-[2-[4-(triphenylmethyl)phenoxy]ethoxy]ethoxy]ethoxy]-3-[2-[2-[2-[4-[2-[2-[4-(triphenylmethyl)phenoxy]ethoxy]ethoxy]ethoxy]phen oxy]ethoxy

CM 1

CRN 145839-35-6 CMF C85 H89 N O12

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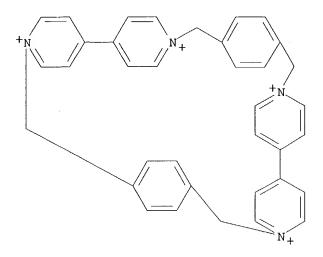
PAGE 1-C

CM 2

CRN 117271-77-9 CMF C36 H32 N4 . 4 F6 P CHU 10/001756 3/17/04 Page 54

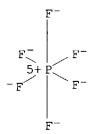
CM 3

CRN 117271-76-8 CMF C36 H32 N4



CM 4

CRN 16919-18-9 CMF F6 P CCI CCS



L33 ANSWER 13 OF 13 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1993:234028 HCAPLUS

DN 118:234028

ED Entered STN: 12 Jun 1993

TI Towards controllable molecular shuttles. 2

AU Ashton, Peter R.; Bissell, Richard A.; Gorski, Romuald; Philp, Douglas; Spencer, Neil; Stoddart, J. Fraser; Tolley, Malcolm S.

CS Sch. Chem., Univ. Birmingham, Edgbaston/Birmingham, B15 2TT, UK

SO Synlett (1992), (11), 919-22 CODEN: SYNLES; ISSN: 0936-5214

DT Journal

LA English

CC 28-23 (Heterocyclic Compounds (More Than One Hetero Atom))

GΙ

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* STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT *
     The design and synthesis of a mol. shuttle I, in which the two components
AB
     are a cyclobis (paraquat-p-phenylene) tetracationic macrocycle and a linear
     polyether chain intercepted by a 2,3,5-trisubstituted indole unit and a
     hydroquinol residue, and terminated by two 4-tritylphenyl ether functions,
     are described. The starting materials for this synthesis were
     4-PhCH2OC6H4XH (X = NH, O), 4-Ph3CC6H4OH, and ClCH2CH2OCH2CH2OH2. A
     key step in the preparation of I is the Fischer indole synthesis of the linear
     component of the mol. shuttle from 4-Ph3C6H4OCH2CH2OCH2CH2OCH2CH2OC6H4NHNH
     CO2CMe3-4 and ketal II.
ST
     mol shuttle controllable; cyclobisparaquatphenylenetetracationic
     macrocycle linear polyether indole; benzyloxyaniline benzyloxyphenol
     tritylphenol conversion mol shuttle
IT
     5197-62-6, 2-[2-(2-Chloroethoxy)ethoxy]ethanol
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (alkylation by, of trityl- and benzyloxyphenol)
IT
     5978-08-5
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (alkylation by, of tritylphenoxyethoxyethoxyethoxyphenoxyethoxyet
        hanol)
TT
     103-16-2, 4-Benzyloxyphenol
                                   978-86-9, 4-Tritylphenol
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (alkylation of, with chloroethoxyethoxyethanol)
IT
     108861-20-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (cyclocondensation of, with bis(bromomethyl)benzene in synthesis of
        mol. shuttle)
ΙT
     623-24-5, 1,4-Bis (bromomethyl) benzene
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (cyclocondensation of, with phenylenebis(methylbipyridinium salt) in
        synthesis of mol. shuttle)
IT
     6373-46-2
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (diazotization and reduction of)
     147553-58-0P 147553-59-1P
IT
     RL: PRP (Properties); PREP (Preparation)
        (formation and spectra of, mol. shuttle synthesis in relation to)
IT
     145839-40-3P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and alkylation by, of hydroxyethoxyethoxyethoxyphenol)
IT
     145839-41-4P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and alkylation of, with chloropentanone ketal)
IT
     127943-23-1P
                   145839-39-0P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and alkylation of, with tritylphenoxyethoxyethoxyethyl
        tosylate)
IT
     51145-58-5P
     RL: SPN (Synthetic preparation); PREP (Preparation)
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(preparation and butoxycarbonylation of)

IT 145839-38-9P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and debenzylation of)

IT 145839-36-7P 145839-37-8P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and reaction of, in Fischer indole synthesis)

İT 147553-57-9P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

IT 145839-35-6P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of, as intermediate in synthesis of mol. shuttle)

IT 145839-35-6P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of, as intermediate in synthesis of mol. shuttle)

RN 145839-35-6 HCAPLUS

CN 1H-Indole, 2-methyl-5-[2-[2-[4-(triphenylmethyl)phenoxy]ethoxy]ethoxy]ethoxy] thoxy]-3-[2-[2-[2-[4-[2-[2-[4-(triphenylmethyl)phenoxy]ethoxy

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